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Surface nitrous oxide concentrations and fluxes from water bodies of the agricultural watershed in Eastern China

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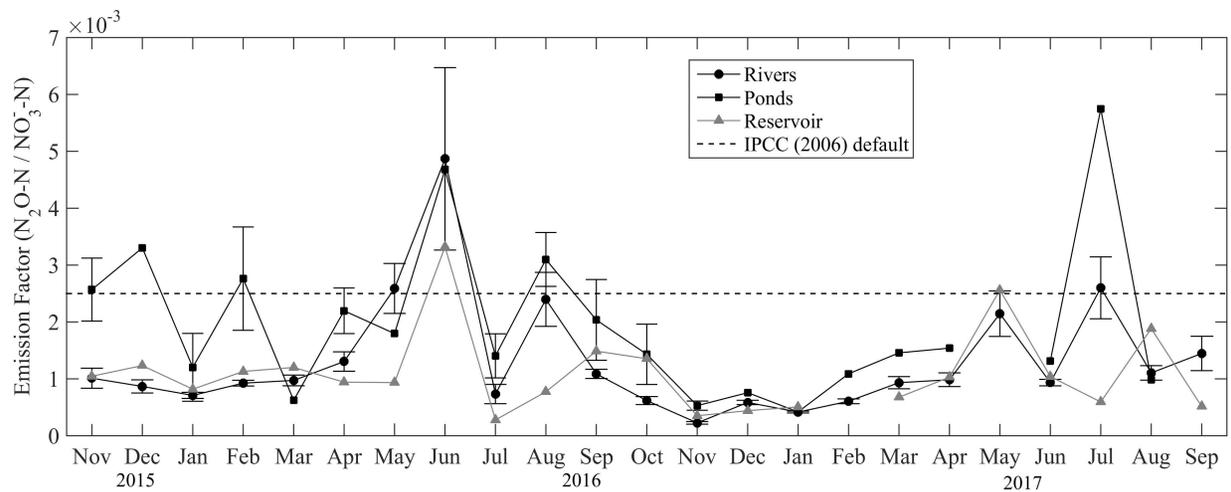
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1 **Surface Nitrous Oxide Concentrations and Fluxes from Water**  
2 **Bodies of the Agricultural Watershed in Eastern China**

3

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24 **Abstract:** Agriculture is one of major emission sources of nitrous oxide ( $\text{N}_2\text{O}$ ), an  
25 important greenhouse gas dominating stratospheric ozone destruction. However,  
26 indirect  $\text{N}_2\text{O}$  emissions from agriculture watershed water surfaces are poorly  
27 understood. Here, surface-dissolved  $\text{N}_2\text{O}$  concentration in water bodies of the  
28 agricultural watershed in Eastern China, one of the most intensive agricultural regions,  
29 was measured over a two-year period. Results showed that the dissolved  $\text{N}_2\text{O}$   
30 concentrations varied in samples taken from different water types, and the annual  
31 mean  $\text{N}_2\text{O}$  concentrations for rivers, ponds, reservoir, and ditches were  $30\pm 18$ ,  $19\pm 7$ ,  
32  $16\pm 5$  and  $58\pm 69$   $\text{nmol L}^{-1}$ , respectively. The  $\text{N}_2\text{O}$  concentrations can be best predicted  
33 by the  $\text{NO}_3^-$ -N concentrations in rivers and by the  $\text{NH}_4^+$ -N concentrations in ponds.  
34 Heavy precipitation induced hot moments of riverine  $\text{N}_2\text{O}$  emissions were observed  
35 during farming season. Upstream waters are hot spots, in which the  $\text{N}_2\text{O}$  production  
36 rates were two times greater than in non-hotspot locations. The modeled watershed  
37 indirect  $\text{N}_2\text{O}$  emission rates were comparable to direct emission from fertilized soil. A  
38 rough estimate suggests that indirect  $\text{N}_2\text{O}$  emissions yield approximately 4% of the  
39 total  $\text{N}_2\text{O}$  emissions yield from N-fertilizer at the watershed scale. Separate emission  
40 factors (EF) established for rivers, ponds, and reservoir were 0.0013, 0.0020, and  
41 0.0012, respectively, indicating that the IPCC (Inter-governmental Panel on Climate  
42 Change) default value of 0.0025 may overestimate the indirect  $\text{N}_2\text{O}$  emission from  
43 surface water in eastern China. EF was inversely correlated with N loading,  
44 highlighting the potential constraints in the IPCC methodology for water with a high  
45 anthropogenic N input.

46

47 **Capsule abstract**48 The surface indirect  $\text{N}_2\text{O}$  emission rates were comparable to direct emission, but the

49 IPCC default emission factor may overestimate indirect N<sub>2</sub>O emission in the region.

50

51 **Keywords:** Indirect emission; N<sub>2</sub>O concentrations; emission rates; emission factor;

52 IPCC

## 53 **Introduction**

54 Nitrous oxide (N<sub>2</sub>O) is a powerful and long-lived greenhouse gas and also a dominant  
55 component in the destruction of the stratospheric ozone (Ravishankara et al. 2009).  
56 The atmospheric N<sub>2</sub>O concentration is estimated to have increased by 22% since 1750,  
57 with a current concentration of 329 ppb (Davidson 2009; Cooper et al. 2017). The  
58 increasing atmospheric N<sub>2</sub>O concentration has received considerable attention. The  
59 estimated annual global N<sub>2</sub>O emission is 13.3~18.8 Tg N yr<sup>-1</sup>, to which agriculture  
60 contributes nearly 30% (Xu et al. 2008; Cooper et al. 2017). Importantly, agriculture  
61 is considered by far the largest source (~80%) of anthropogenic N<sub>2</sub>O via both direct  
62 emission and indirect emission (Kroeze et al. 1999; Davidson 2009; Cooper et al.  
63 2017).

64  
65 Anthropogenic N<sub>2</sub>O input into the atmosphere from fertilized soils are defined as  
66 direct emission, while indirect emission represents N<sub>2</sub>O production in ditches, streams,  
67 and rivers induced by leaching and runoff of reactive N from agricultural areas  
68 (Mosier et al. 1998; Fu et al. 2018). Direct N<sub>2</sub>O emissions have been well documented  
69 through a large amount of field measurements (Xu et al. 2008; Shcherbak et al. 2014),  
70 while indirect emissions from agricultural watershed are less well constrained due to  
71 the scarcity of studies on this topic (Reay et al. 2003; Outram and Hiscock; Fu et al.  
72 2018). Importantly, indirect emission is estimated to account for over one-quarter of  
73 global total agricultural N<sub>2</sub>O emissions (Reay et al. 2012; Venkiteswaran et al. 2014)  
74 and dominates the inter-annual variability of total emission (Griffis et al. 2017).

75 Meanwhile, riverine N<sub>2</sub>O emissions, especially in agricultural areas, remain a major  
76 source of uncertainty in the global N<sub>2</sub>O budget due to the large spatiotemporal  
77 variability (Beaulieu et al. 2011; Borges et al. 2015; Audet et al. 2017). More field  
78 measurements of indirect N<sub>2</sub>O emission are needed to reduce the uncertainty and  
79 improve the reliability of the global anthropogenic N<sub>2</sub>O budget (Reay et al. 2012;  
80 Saikawa et al. 2014).

81

82 IPCC provided emission factors (EF) to estimate indirect N<sub>2</sub>O emissions resulting  
83 from fertilizer-N and manure-N loss via leaching and runoff (De Klein et al. 2006), in  
84 which global and regional N<sub>2</sub>O emissions from a water body can be calculated by  
85 multiplying the nitrogen fertilizer input or relevant anthropogenic N loading with the  
86 default EF (Hu et al. 2016; Hama-Aziz et al. 2017; Fu et al. 2018). Since 2006, the  
87 IPCC default EF values for rivers, groundwater, and estuaries were both set to 0.0025  
88 (De Klein et al., 2006). However, studies have shown that the default EF is poorly  
89 constrained due to high variability in environmental conditions (Turner et al. 2015;  
90 Cooper et al. 2017; Griffis et al. 2017). For example, a field measurement showed that  
91 the IPCC default EF is underestimated up to nine-fold in rivers in the U.S. Corn Belt,  
92 an intensively agricultural region (Turner et al. 2015; Fu et al. 2017; Fu et al. 2018),  
93 but it is unexpectedly overestimated in an intensively agriculture area in other regions  
94 (Clough et al. 2007; Outram and Hiscock 2012; Hama-Aziz et al. 2017). Using the  
95 default IPCC emission factor for predicting indirect N<sub>2</sub>O emissions from all rivers  
96 may be inappropriate, and more measurements are needed to accurately calculate the

97  $EF_{5r}$  (Outram and Hiscock 2012; Hama-Aziz et al. 2017; Tian et al. 2017).

98

99 Heavy farmland N fertilizer application in Eastern China, one of most intensively  
100 agricultural regions in the world, has led to a high N loading input in the watershed.  
101 The fertilizer N application rate is up to  $600 \text{ kg N ha}^{-1} \text{ yr}^{-1}$  in Eastern China (Ju et al.  
102 2009; Xing and Zhu 2002). For comparison, the value is only approximately  $100 \text{ kg N}$   
103  $\text{ha}^{-1} \text{ yr}^{-1}$  in intensively agricultural regions in the US (Griffis et al. 2017; Turner et al.  
104 2015). Further, a large amount of farmland N-fertilizer ( $\sim 280 \text{ kg N ha}^{-1} \text{ yr}^{-1}$ ) enters the  
105 watershed via runoff and leaching in Eastern China (Xing and Zhu 2002; Yan et al.  
106 2011), which may significantly increase the  $\text{N}_2\text{O}$  production rates (Beaulieu et al.  
107 2011). Previous studies suggested that the surface water  $\text{N}_2\text{O}$  emissions in Eastern  
108 China may be very high due to the vast farmland N-fertilizer (Seitzinger and Kroeze  
109 1998). However, until now, limited information is available regarding the  
110 characteristics of watershed  $\text{N}_2\text{O}$  dynamics in this area.

111

112 In this study, we measured the dissolved  $\text{N}_2\text{O}$  concentration in water bodies in a  
113 typical agricultural watershed in Eastern China. Our objectives were to quantify the  
114 importance of agricultural watershed fueled by farmland N-fertilizer as sources of  
115 indirect  $\text{N}_2\text{O}$  emission in this area, to explore the indirect  $\text{N}_2\text{O}$  emission factor, and to  
116 further investigate the spatiotemporal characteristics of  $\text{N}_2\text{O}$  dynamics.

117

118 **2 Materials and Methods**

## 119 **2.1 Study area**

120 This field measurement was carried out in a typical agriculture watershed located in  
121 Eastern China (31°58' to 32°01'N, 119°12' to 119°14'E; Figure S1). The watershed  
122 has previously been reported in detail (Yan et al. 2011). Briefly, there is a reservoir,  
123 three rivers, and thousands of small pond in the watershed. The average annual  
124 precipitation is 1100 mm and the mean annual temperature is 15 °C. The major annual  
125 cropping rotations are rice-wheat for the paddy fields and maize-rapeseed,  
126 respectively. There is no industry in the studied area, and agriculture is the dominant  
127 local source of anthropogenic N.

128

## 129 **2.2 Sample collection and analysis**

130 A watershed-scale sampling round was conducted in each month from October 2015  
131 to September 2017, during which samples were taken from the surface water of the  
132 reservoir, rivers, ponds, and ditches (Figure S1). As a whole, water samples were  
133 collected from 30 sampling sites, of which 1 site was in a reservoir, 19 sites were in  
134 rivers, 3 sites were in ponds, and 7 sites were in ditches. There are three rivers in the  
135 watershed: River 1, River 2, and River 3, respectively. Given the river length, the  
136 sampling sites were divided into two sections in River 1 (midstream and downstream)  
137 and three sections (upstream, midstream, and downstream) in River 2, respectively. It  
138 should be noted that the ditches sometimes dry up without covering water, leading to  
139 discontinuous sampling.

140

141 Each watershed-scale survey was completed over two consecutive days. The  
142 procedures of sampling and analysis have previously been reported in detail in Xiao et  
143 al. 2019. Briefly, triplicate bubble-free water samples were collected at a 20-cm depth  
144 below the surface using 300 mL glass bottles, and these bottles were transported to lab  
145 for dissolved N<sub>2</sub>O concentration analysis using the headspace equilibration method.  
146 Water samples were also collected using an organic glass hydrophore to the  
147 ammonium-nitrogen (NH<sub>4</sub><sup>+</sup>-N), nitrate-nitrogen (NO<sub>3</sub><sup>-</sup>-N), and nitrite-nitrogen (NO<sub>2</sub><sup>-</sup>-N)  
148 analysis by a continuous flow analyzer (Skalar san<sup>++</sup>, Netherlands). The water  
149 temperature, pH, specific conductance (SpC), and dissolved oxygen concentration  
150 (DO) were measured *in situ* with a multiparameter probe (YSI 650MDS, YSI Inc.  
151 Yellow Springs, OH, USA) at a 20-cm depth to be consistent with water sampling.

#### 153 2.4 N<sub>2</sub>O fluxes and emission factor calculations

154 The N<sub>2</sub>O fluxes ( $F_n$ ,  $\mu\text{mol m}^{-2} \text{d}^{-1}$ , positive indications of N<sub>2</sub>O emission from water to  
155 atmosphere) was calculated using water-air gas exchange model:

$$156 \quad F_n = k \times (C_w - C_{eq}) \quad (1)$$

157 where  $C_w$  is the surface dissolved N<sub>2</sub>O concentration ( $\mu\text{mol m}^{-3}$ ) in water measured by  
158 headspace equilibration method,  $C_{eq}$  is the N<sub>2</sub>O concentration in water that is in  
159 equilibrium with the atmosphere at the *in-situ* temperature, and  $k$  is the gas transfer  
160 coefficient. For the rivers,  $k$  was calculated considering both wind and water  
161 turbulence (Clough et al. 2007):

$$162 \quad k = 2.78E^{-6} au_{10}^2 \left(\frac{S_c}{660}\right)^{0.5} + \sqrt{\frac{DU}{h}} \quad (2)$$

163 where  $2.78E^{-6}$  is a conversion factor ( $\text{cm h}^{-1}$  to  $\text{m s}^{-1}$ ),  $a$  is a constant (0.31),  $u_{10}$  is the  
164 wind speed at a height of 10 m above the water surface, and  $S_c$  is the Schmidt number  
165 for  $\text{N}_2\text{O}$ ,  $U$  is the river water velocity ( $\text{m s}^{-1}$ ),  $h$  is the average river depth (m), and  $D$   
166 is the  $\text{N}_2\text{O}$  diffusion coefficient in the water. The hourly measurement of wind speed  
167 was obtained from a weather station in the watershed to calculate the  $k$ . The river  
168 water velocity is steady and only varies depending upon precipitation (Yan et al. 2011;  
169 Xia et al. 2013b).

170

171 For the lentic ecosystem (reservoir and ponds), the  $\text{N}_2\text{O}$  transfer velocity is mainly  
172 driven by wind speed because no surface water flow occurs.  $k$  ( $\text{cm h}^{-1}$ ) was calculated  
173 from a wind-dependent formula derived from a small shallow lake as follows (Cole  
174 and Caraco 1998):

$$175 \quad k = (S_c/660)^{-n} (2.07 + 0.215 \times u_{10}^{1.7}) \quad (3)$$

176 There is no separate equation for  $k$  calculation for ditches, and Equation (3) was also  
177 used for calculating the gas transfer coefficient  $k$  for ditches.

178

179 The indirect  $\text{N}_2\text{O}$  emission factor (EF) for rivers, ponds, ditches, and the reservoir  
180 were calculated using the common IPCC methodology as follows (De Klein et al.  
181 2006):

$$182 \quad \text{EF} = \text{N}_2\text{O-N}/\text{NO}_3^- \text{-N} \quad (4)$$

183 Where  $\text{N}_2\text{O-N}$  and  $\text{NO}_3^- \text{-N}$  are concentrations measured in rivers, ponds, ditches, and  
184 the reservoir. Equation (4) has been widely used to calculate the indirect  $\text{N}_2\text{O}$

185 emission factor (Turner et al. 2015; Hama-Aziz et al. 2017; Tian et al. 2017).

186

## 187 **2.5 Data analysis**

188 Statistical analyses were conducted with SPSS version 18.0 (SPSS, Inc., USA).

189 In the study area, spring is from March to the end of May, summer is from June to

190 August, autumn is from September to November, and winter is from December to

191 February in next year. Simple linear and multi-linear stepwise regression analyses

192 were performed to determine the relationships among  $N_2O$ , the emission factor, and

193 independent variables (e.g.,  $NH_4^+$ -N,  $NO_3^-$ -N, and  $NO_2^-$ -N). The entire two-year

194 dataset was divided according to water type (e.g., rivers, ponds, ditches, and reservoir),

195 and the difference of their mean  $N_2O$  concentrations and fluxes was examined using a

196 one-way analysis of variance (ANOVA). After completing the ANOVA test, a least

197 significant difference (LSD) post hoc test was conducted. Differences at  $p < 0.05$  were

198 labeled as statistically significant in all analyses.

199

## 200 **3 Results**

### 201 **3.1 Physical and chemical characteristics**

202 There was no obvious variation in water temperature across different water body

203 types in the agricultural watershed (Figure 2S(a)). The annual mean water temperature

204 is 19.3 °C over the two-year study period. There was a clear seasonality for water

205 temperature, with the highest water temperature appearing in the summer (33.3 °C)

206 and the lowest (6.3 °C) in winter. The annual mean wind speed was 2.5 m s<sup>-1</sup> based on

207 the measurement from the local weather station.

208

209 In contrast to water temperature, the surface water chemical properties varied among  
210 rivers, ponds, reservoir, and ditches (Table 1 and Figure 2S). Generally, the  $\text{NO}_3^-$ -N  
211 concentrations was highest in ditches with annual mean value of  $1.85 \text{ mg L}^{-1}$ ,  
212 followed by rivers ( $0.99 \text{ mg L}^{-1}$ ), reservoir ( $0.54 \text{ mg L}^{-1}$ ) and ponds ( $0.47 \text{ mg L}^{-1}$ ). In  
213 contrast, the highest  $\text{NH}_4^+$ -N concentration appeared in ponds with annual mean value  
214 of  $0.32 \text{ mg L}^{-1}$ , followed by rivers and reservoirs. The  $\text{NO}_2^-$ -N concentrations in these  
215 surface water areas showed minor differences. The average DO concentration was  
216 over-saturated only in the reservoir with a value of  $10.45 \text{ mg L}^{-1}$  or 110% saturation,  
217 and the annual mean values for rivers, ponds, and ditches were 6.65, 5.78, and 7.90  
218  $\text{mg L}^{-1}$ , respectively.

219

220 The  $\text{NO}_3^-$ -N was the dominant form of the total dissolved inorganic nitrogen (DIN) in  
221 the surface water of the agricultural watershed (Table 1). For example, the  $\text{NO}_3^-$ -N  
222 concentration accounted for 76%, 57%, 72%, and 92% of the total DIN concentration  
223 for rivers, ponds, the reservoir, and ditches, respectively. In addition, the surface water  
224  $\text{NO}_3^-$ -N concentrations for the rivers, ponds, and reservoir showed clear seasonality  
225 with peak concentrations being observed in summer and winter (Figure 2S(b)),  
226 coincides with the crop-planting period. There were no obvious temporal variations  
227 for  $\text{NH}_4^+$ -N concentration except in ponds.

228

### 229 3.2 Spatial and temporal characteristics of surface water N<sub>2</sub>O concentration

230 The dissolved N<sub>2</sub>O concentration in surface water varied across water body types  
231 (Figure S3). Dissolved N<sub>2</sub>O concentrations in the ditches spanned a broad range, from  
232 10 to 243 nmol L<sup>-1</sup>, but varied within relatively narrow ranges of 4-172 nmol L<sup>-1</sup>, 5-50  
233 nmol L<sup>-1</sup>, and 9-30 nmol L<sup>-1</sup> in rivers, pond, and reservoir, respectively. Meanwhile,  
234 the annual mean surface N<sub>2</sub>O concentration in ditches with a value of 63 nmol L<sup>-1</sup> was  
235 significantly ( $p < 0.01$ ) higher than those in rivers (29 nmol L<sup>-1</sup>), ponds (20 nmol L<sup>-1</sup>),  
236 and reservoir (15 nmol L<sup>-1</sup>). The differences of N<sub>2</sub>O concentrations between rivers,  
237 ponds, and reservoir were insignificant ( $p > 0.05$ ).

238

239 The N<sub>2</sub>O concentrations also varied within rivers. River 3 with the shortest transport  
240 length (Figure S1) had the highest N<sub>2</sub>O concentration compared to River 1 and River  
241 2. The overall mean N<sub>2</sub>O concentrations were 26, 30, and 35 for rivers 1, 2, and 3,  
242 respectively. Further analysis showed large spatial variations for the riverine N<sub>2</sub>O  
243 concentration along the current direction of the water (Figure S4). A N<sub>2</sub>O  
244 concentration gradient from the upstream to the estuary is clearly evident. Peak N<sub>2</sub>O  
245 concentrations were typically found in the upstream of the river. The annual mean  
246 N<sub>2</sub>O concentrations in upstream, midstream, and downstream of River 2 were 45, 25,  
247 and 21 nmol L<sup>-1</sup>, respectively. River 3 is too short to be sampled at different locations,  
248 and samples were only taken from midstream and downstream in River 1, but the  
249 measured data also showed that the N<sub>2</sub>O concentrations decreased rapidly far from the  
250 upstream areas (Figure 4S(b)).

251

252 Another notable feature was that the surface N<sub>2</sub>O concentration varied over time,  
253 especially in rivers. Hot moments, or rapid temporal increases in the surface N<sub>2</sub>O  
254 concentrations, were evident in rivers (Figure 1). There was clear seasonality for the  
255 monthly mean N<sub>2</sub>O concentration in rivers, ranging from 15 to 95 nmol L<sup>-1</sup>, with peak  
256 concentrations being observed in the summer. In contrast to rivers, the temporal  
257 variation showed a narrower range for ponds (7-34 nmol L<sup>-1</sup>) and the reservoir (9-30  
258 nmol L<sup>-1</sup>) but with peak concentrations being observed mostly in the winter. For rivers,  
259 the summertime N<sub>2</sub>O concentrations with a mean value of 48 nmol L<sup>-1</sup> were  
260 significantly ( $p < 0.05$ ) higher than those in spring (19 nmol L<sup>-1</sup>), autumn (24 nmol  
261 L<sup>-1</sup>), and winter (28 nmol L<sup>-1</sup>). For ponds, the summertime N<sub>2</sub>O concentration with a  
262 mean value of 16 nmol L<sup>-1</sup> was significant ( $p < 0.05$ ) lower than that in winter (25  
263 nmol L<sup>-1</sup>). Additionally, the seasonal average values in the reservoir were 13 (spring),  
264 18 (summer), 14 (autumn), and 18 nmol L<sup>-1</sup> (winter), respectively, showing significant  
265 ( $p = 0.046$ ) difference between spring and summer. We cannot characterize the  
266 seasonal variation of N<sub>2</sub>O concentrations in ditches due to non-continuous  
267 measurements in time.

268

### 269 **3.3 Relationships between N<sub>2</sub>O concentration and environment factors**

270 Surface N<sub>2</sub>O concentration in the rivers, ponds, and reservoir exhibited different  
271 responses to inorganic N loadings (Figure S5). The N<sub>2</sub>O concentration and NO<sub>3</sub><sup>-</sup>-N  
272 concentration exhibited a stronger correlation in rivers ( $r = 0.71$ ,  $p < 0.01$ ) than in

273 ponds ( $r = 0.38$ ,  $p = 0.03$ ). Inversely, a stronger correlation between  $N_2O$   
274 concentration and  $NH_4^+$ -N concentration was found in ponds ( $r = 0.65$ ,  $p < 0.01$ ) than  
275 in rivers ( $r = 0.33$ ,  $p < 0.01$ ). Additionally, the  $NO_2^-$ -N concentration was well  
276 correlated with the riverine  $N_2O$  concentration ( $r = 0.65$ ,  $p < 0.01$ ). In the reservoir,  
277 the  $N_2O$  concentrations showed no significant correlation with any form of inorganic  
278 N loadings.

279

280 The temporal correlation between the surface water  $N_2O$  concentration and water  
281 temperature were insignificant for rivers ( $r = 0.27$ ,  $p = 0.20$ ), ponds ( $r = 0.37$ ,  $p =$   
282  $0.08$ ), and reservoir ( $r = 0.11$ ,  $p = 0.61$ ). Correlation analyses also revealed that the  
283 temporal variation in  $N_2O$  concentration was not correlated well with the water  
284 chemical properties (e.g., Spc, pH, and DO). Precipitation, by contrast, appeared to  
285 contribute to the temporal variation of the  $N_2O$  concentration (Figure 2). The  
286 accumulated precipitation for ten days before sampling was well correlated with the  
287 monthly riverine  $N_2O$  concentration when the precipitation amounts were greater than  
288 50 mm ( $r = 0.92$ ,  $p < 0.01$ ). Importantly, the hot moments, the rapid temporal increase  
289 in the riverine  $N_2O$  production (Fu et al. 2018), were induced by heavy precipitation  
290 during the crop-planting period in the study (Figure 1). However, the temporal  
291 correlation between precipitation and the  $N_2O$  concentration in the lentic ecosystems  
292 (ponds and reservoir) was insignificant ( $p > 0.05$ ).

293

294 **3.4 Indirect  $N_2O$  fluxes**

295 The mean estimated indirect N<sub>2</sub>O emission (mean ± 1 standard deviation) from the  
296 surface water of the agricultural watershed was 27.1±23.4 μmol m<sup>-2</sup> d<sup>-1</sup> based on two  
297 years' field measurements. The rivers were found to have the highest N<sub>2</sub>O fluxes,  
298 with a mean value of 49.6 μmol m<sup>-2</sup> d<sup>-1</sup>. Ditches had the second highest N<sub>2</sub>O fluxes  
299 with large uncertainty (mean ± 1 standard deviation: 45.2±95.1 μmol m<sup>-2</sup> d<sup>-1</sup>; Table 1).  
300 The ponds and reservoir showed moderate emission fluxes with a mean value of 8.0  
301 and 5.6 μmol m<sup>-2</sup> d<sup>-1</sup>, respectively. The highest N<sub>2</sub>O fluxes for the three water body  
302 types — rivers, ponds, and reservoir — were found in the summer (Table 1).

303

304 The indirect N<sub>2</sub>O fluxes from water bodies in the watershed were calculated with the  
305 dissolved N<sub>2</sub>O concentration. The calculated indirect N<sub>2</sub>O fluxes and dissolved N<sub>2</sub>O  
306 concentration were highly correlated for all records across water body types and over  
307 time ( $r = 0.97, p < 0.01$ ). The roles of environment factors influencing the N<sub>2</sub>O fluxes  
308 were similar to those of surface water-dissolved N<sub>2</sub>O concentration.

309

### 310 **3.5 Indirect N<sub>2</sub>O emission factor**

311 The results revealed that the EF was not uniform for the different water types (Figure  
312 3). Ponds had the highest EF, followed by ditches and rivers on average (Table 1). The  
313 reservoir, the large water body in the watershed, had the lowest EF with a mean value  
314 of 0.0011. The EF of 0.0013 for rivers was nearly half of the default 0.0025 by IPCC.  
315 Additionally, the measured EF of 0.0014 for ditches was also significantly lower than  
316 the IPCC default value of 0.0025 for river.

317

318 It was evidenced from the field measurements that EF varied over time (Figure 3).

319 The monthly mean EF varied from 0.0002 to 0.0049 for rivers, 0.0004 to 0.0057 for

320 ponds, and 0.0002 to 0.0033 for the reservoir over the two-year period. The riverine

321 EF showed apparent seasonality: summer (mean  $\pm$  1 standard deviation:

322  $0.0021 \pm 0.0016$ ) > spring ( $0.0015 \pm 0.0007$ ) > autumn ( $0.0009 \pm 0.0005$ ) > winter

323 ( $0.0007 \pm 0.0002$ , Table 1). In contrast to  $N_2O$  concentration, the EFs were highly

324 negative with the  $NO_3^-$ -N concentration for rivers, ponds, and the reservoir (Figure 4).

325

## 326 **4 Discussion**

### 327 **4.1 Controls on $N_2O$ variability**

328 Previous field measurements demonstrated that although other environmental factors

329 such as dissolved oxygen (Kampschreur et al. 2009; Rosamond et al. 2012), organic

330 carbon (Kampschreur et al. 2009), and water temperature (Venkiteswaran et al. 2014)

331 affect the riverine  $N_2O$  production rate, the  $NO_3^-$ -N concentration was the

332 predominant control on the riverine  $N_2O$  concentration in this study (Figure S5(a)).

333 The pattern was consistent with previous observations, showing that  $NO_3^-$ -N is an

334 important driver of riverine  $N_2O$  production in agricultural watersheds (Beaulieu et al.

335 2008; Baulch et al. 2011; Turner et al. 2016; Hama-Aziz et al. 2017; Tian et al. 2017).

336 It is likely that farmland N fertilizer loading to rivers stimulates denitrification and

337 associated  $N_2O$  production (Kampschreur et al. 2009; Beaulieu et al. 2011; Maavara

338 et al. 2019).  $NH_4^+$ -N concentrations, a good indicator of domestic pollution (Garnier

339 et al. 2009), were found to peak in ponds (Table 1). Additionally, the strong  
340 relationship between  $\text{NH}_4^+\text{-N}$  and pond  $\text{N}_2\text{O}$  concentration ( $r = 0.65$ ,  $p < 0.01$ ; Figure  
341 S5(b)) is reasonable because the high  $\text{NH}_4^+\text{-N}$  level may increase the  $\text{N}_2\text{O}$   
342 accumulation via nitrification, which had been demonstrated in high population  
343 density areas (Xia et al. 2013a; Yu et al. 2013; Hu et al. 2016). The poor correlation  
344 between  $\text{N}_2\text{O}$  and any form of N loading in the reservoir, by contrast, indicated that  
345 other biotic and abiotic factors played a greater role than N loading in  $\text{N}_2\text{O}$  production  
346 (Deemer et al. 2016).

347

348 Both field measurement and model simulation find that the riverine  $\text{N}_2\text{O}$  production  
349 rate is highly sensitive to perturbations in precipitation (Chen et al. 2016; Cooper et al.  
350 2017; Griffis et al. 2017; Fu et al. 2018). Here, precipitation (10-day accumulated  
351 precipitation prior to sampling  $> 50$  mm) was found to drive the temporal variation of  
352 the riverine  $\text{N}_2\text{O}$  concentration (Figure 2a), consistent with previous studies in the  
353 agriculture watersheds in Southwestern China (Tian et al. 2017) and UK (Cooper et al.  
354 2017). Heavy precipitation is likely to transport more agricultural N loading to the  
355 river (Sinha et al. 2017; Figure 2b) and subsequently stimulates  $\text{N}_2\text{O}$  production (Liu  
356 and Greaver 2009; Tian et al. 2017). The results reported here showed that heavy  
357 precipitation combined with intense agricultural fertilizer application during the  
358 farming season triggered riverine  $\text{N}_2\text{O}$  hot moments (Figure 1), which should be  
359 captured to reduce the uncertainty of the  $\text{N}_2\text{O}$  budget (Fu et al. 2018; Shrestha and  
360 Wang 2018). The projected increase in both heavy and total precipitation induced by

361 climate change (Sinha et al. 2017), together with the increase in the use of nitrogen  
362 fertilizer to meet the food demand (Jiang et al. 2010; Saikawa et al. 2014), posed  
363 significant potential for agricultural watershed indirect N<sub>2</sub>O production in the global  
364 N<sub>2</sub>O budget.

365

366 Although many previous studies showed a strong positive correlation between  
367 temperature and riverine N<sub>2</sub>O production (Beaulieu et al. 2010; Hinshaw and  
368 Dahlgren 2012; Shrestha and Wang 2018), our results reported here were consistent  
369 with the field measurement in a Swedish agricultural catchment showing water  
370 temperature had an insignificant effect on seasonal variation of N<sub>2</sub>O concentration  
371 (Audet et al. 2017). This is probably due to the N concentration outweighs the  
372 temperature in determining the water N<sub>2</sub>O production via nitrification and  
373 denitrification (Kampschreur et al. 2009; Davidson et al. 2015; Capodici et al. 2018).

374 Another notable feature was that the surface NO<sub>3</sub><sup>-</sup>-N concentration in rivers was the  
375 highest in winter, but the highest N<sub>2</sub>O concentration occurred in summer in this study  
376 (Table 1). Some studies have proposed that the carbon limitation during denitrification  
377 is associated with increased N<sub>2</sub>O production (Kampschreur et al. 2009; Capodici et al.  
378 2018), which may be the explanation for this pattern. Because study has shown that  
379 the dissolved organic carbon in these rivers increased from 6.2 mg L<sup>-1</sup> in summer to  
380 18.6 mg L<sup>-1</sup> in winter, with the C:N ratio increased from 1.3 to 3.1 (Zhao et al. 2013).

381 The high C:N ratio may supply availability carbon sources, resulting in an increase in  
382 denitrification efficiency and associated decrease in N<sub>2</sub>O production although the high

383  $\text{NO}_3^-$ -N concentration in winter in this study (Kampschreur et al. 2009; Zhao et al.  
384 2013; Capodici et al. 2018).

385

386 The riverine  $\text{N}_2\text{O}$  production rate varies greatly by geographic location (Figure S4).  
387 The most notable feature was that the  $\text{N}_2\text{O}$  concentration decreased sharply from the  
388 upstream to the estuary, with the mean value decreasing from 47 to 17  $\text{nmol L}^{-1}$  in  
389 River 2. The result reported here is consistent with the field measurement in other  
390 intensive agricultural activity areas, such as the Seine basin in France (Garnier et al.  
391 2009), the U.S. Corn Belt (Turner et al. 2015), and a typical agriculture watershed in  
392 the UK (Reay et al. 2003). The high  $\text{NO}_3^-$ -N concentrations in upstream waters  
393 (Figure 4S(a)), may enhance the microbial processes of  $\text{N}_2\text{O}$  production (Herrman et  
394 al. 2008; Zhao et al. 2014). Notably, direct  $\text{N}_2\text{O}$  input from other sources, such as soil  
395 and groundwater, might also lead to a high  $\text{N}_2\text{O}$  concentration in upstream waters  
396 (Garnier et al. 2009; Beaulieu et al. 2011). As suggested in previous studies, the  
397 upstream waters are the hot spot for  $\text{N}_2\text{O}$  production (Peterson et al. 2001; Turner et al.  
398 2015). In summary,  $\text{N}_2\text{O}$  production rates in hotspots were two times greater than in  
399 non-hotspot locations in the presented study.

400

#### 401 **4.2 Indirect $\text{N}_2\text{O}$ fluxes in the agricultural watershed**

402 The combined indirect  $\text{N}_2\text{O}$  fluxes from rivers, ponds, and reservoir were calculated  
403 to be 383  $\text{kg N}_2\text{O-N yr}^{-1}$ . The rivers were the most significant contributor of  $\text{N}_2\text{O}$   
404 fluxes of all water bodies, contributing to a total of 161  $\text{kg N}_2\text{O-N yr}^{-1}$ . Total direct

405 N<sub>2</sub>O fluxes from the watershed fertilized soil was estimated to be 12 t (10<sup>3</sup> kg) N yr<sup>-1</sup>  
406 in the watershed (Xia et al. 2013b). Based on a study in another similar region in  
407 Eastern China (She et al. 2018), we roughly estimated that the ditch area for the  
408 rice-wheat field is 25 hm<sup>2</sup>. Adding the annual N<sub>2</sub>O emission yield from agriculture  
409 ditches (~ 117 kg N<sub>2</sub>O-N yr<sup>-1</sup>) in the watershed, the watershed indirect N<sub>2</sub>O emission  
410 thus represents approximately 4% of the total N<sub>2</sub>O emissions from N-fertilizer.  
411 Although the obtained percentage was similar to other studies (Beaulieu et al. 2008;  
412 Garnier et al. 2009; Audet et al. 2017), caution should be taken for the following  
413 reasons: (1) the ditch areas and N<sub>2</sub>O emission rates (45.2±95.1 μmol m<sup>-2</sup> d<sup>-1</sup>) had large  
414 uncertainty; (2) the presented sampling strategies might omit some riverine emission  
415 hot spots and hot moments, which may lead to low biases in the annual total N<sub>2</sub>O  
416 emission yield (Fu et al. 2018).

417  
418 For lentic ecosystems, the annual mean N<sub>2</sub>O emission rates in ponds and reservoir  
419 (8.0 and 5.6 μmol m<sup>-2</sup> d<sup>-1</sup>, respectively) were higher than the median emission rate  
420 (3.9 μmol m<sup>-2</sup> d<sup>-1</sup>) in global lakes/reservoirs according to Hu *et al.* (2016). Meanwhile,  
421 these values were also higher than those in temperate lakes with a mean value of 3.4  
422 μmol m<sup>-2</sup> d<sup>-1</sup> (Soued et al. 2016). A heavily polluted lake nearby, Lake Taihu, showed  
423 a moderate N<sub>2</sub>O emission of 3.5 μmol m<sup>-2</sup> d<sup>-1</sup> (Xiao et al. 2019). The gas transfer  
424 velocity of a lentic ecosystem in the study region can be accurately estimated using  
425 the wind-dependent formula, which had been demonstrated previously (Xiao et al.  
426 2017). These results together indicated watershed N inputs fueled the pond and

427 reservoir N<sub>2</sub>O emission.

428

429 The indirect riverine N<sub>2</sub>O fluxes reported here is compared to the results reported by  
430 previous studies (Table S1). The literature review in Table S1 showed that the riverine  
431 N<sub>2</sub>O fluxes across multiple land-use types ranged from 6.7 to 121.3  $\mu\text{mol m}^{-2} \text{d}^{-1}$ , with  
432 a mean value of 51.3  $\mu\text{mol m}^{-2} \text{d}^{-1}$ . Based on a meta-analysis with 169 observations,  
433 the global median riverine N<sub>2</sub>O emission fluxes is 14.4  $\mu\text{mol m}^{-2} \text{d}^{-1}$ , which is  
434 significantly lower than the annual mean flux reported here (49.6  $\mu\text{mol m}^{-2} \text{d}^{-1}$ ). For  
435 comparison, the direct N<sub>2</sub>O emission rate from fertilized soil in the study area is  
436 approximately 60~90  $\mu\text{mol m}^{-2} \text{d}^{-1}$  (Zou et al. 2005; Zhou et al. 2014; Liu et al. 2016).  
437 Therefore, indirect N<sub>2</sub>O emissions caused by N leaching and surface runoff N losses  
438 in agriculture watershed could contribute significantly to the total N<sub>2</sub>O emission.

439

440 The N<sub>2</sub>O emissions from water are often determined by water-air gas exchange model  
441 or direct field measurement such as using floating chamber technique (Outram and  
442 Hiscock 2012; Zhu et al. 2015; Hama-Aziz et al. 2017; Wu et al. 2018; Capodici et al.  
443 2018). The water-air gas exchange model has been frequently employed to quality  
444 N<sub>2</sub>O emission flux from water, accounting for 80% of flux measurements (Hu et al.  
445 2016; Wu et al. 2018), however, the model-based method may either overestimate or  
446 underestimate N<sub>2</sub>O emission due to the uncertainty in gas transfer coefficient (Yu et al.  
447 2013; Raymond et al. 2013; Wu et al. 2018). For comparison, the direct measurement  
448 using floating chamber method showed the N<sub>2</sub>O emission flux were 48.1-66.1  $\mu\text{mol}$

449  $\text{m}^{-2} \text{d}^{-1}$  and below  $8.6 \mu\text{mol m}^{-2} \text{d}^{-1}$  from the rivers and ponds, respectively, within the  
450 same watershed (Li et al. 2011; Xia et al. 2013a; Han et al. 2014), which were similar  
451 to the model-estimated  $\text{N}_2\text{O}$  emission flux in the presented study (river:  $49.6 \mu\text{mol m}^{-2}$   
452  $\text{d}^{-1}$ ; ponds:  $8.0 \mu\text{mol m}^{-2} \text{d}^{-1}$ ). Additionally, previous study has shown that the  $\text{N}_2\text{O}$   
453 emission flux measured by the two methods above exhibits the same seasonal  
454 variation in an agricultural watershed in southeast China (Wu et al. 2018). Both of  
455 these indicate the model-based method was reasonable for quantifying the watershed  
456 indirect  $\text{N}_2\text{O}$  emission in this study.

457

#### 458 **4.3 Implication of the measured EF values**

459 This study revealed that indirect  $\text{N}_2\text{O}$  EFs were not uniform for the different surface  
460 water types within the same watershed (Table 1 and Figure 3). This is consistent with  
461 the field measurement in UK agriculture watershed, which shows that different water  
462 types produce various amounts of  $\text{N}_2\text{O}$  (Outram and Hiscock 2012; Hama-Aziz et al.  
463 2017). Therefore, separate EFs for each water compartment are required to better  
464 understand the indirect  $\text{N}_2\text{O}$  budget. Emission factor for lentic ponds and the reservoir  
465 should also be included because our results showed that ignoring the indirect  $\text{N}_2\text{O}$   
466 emission from the lentic ecosystems would lead to an underestimation of the indirect  
467 emission yield by 60%.

468

469 For rivers and ditches, the EF values of 0.0013 and 0.0014 were both lower than the  
470 IPCC EF value of 0.0025. These results indicated that the IPCC default value of

471 0.0025 may overestimate the agriculture watershed indirect N<sub>2</sub>O emission in the study  
472 region. The recent study by Maavara et al. 2019 using mechanistic modeling approach  
473 also showed IPCC EFs are likely overestimated. Meanwhile, our data showed that the  
474 EF value across different water types varied over time with a peak appearing in the  
475 summer (Figure 3), which is consistent with a previous study (Hama-Aziz et al. 2017).  
476 Thus, it may be more applicable to have more measurements across different seasons.  
477 Meanwhile, we also found that the IPCC approach of using one EF for all rivers may  
478 be inappropriate because EFs were highly variable across rivers, as suggested by  
479 Beaulieu et al. (2008). The EF values for rivers 1, 2, and 3 were 0.0015, 0.0011, and  
480 0.0016, respectively. Since there are no IPCC EFs for the ponds and reservoir  
481 (Outram and Hiscock 2012), a comparison cannot be made. However, a synthesis  
482 analysis showed that EF for ponds and reservoirs was 0.0012 (Tian et al. 2018), which  
483 was significantly lower than that reported EF value of pond (0.0020) but was same as  
484 the value of reservoir in this study.

485

486 Compared to the previous results, many studies have also observed lower EFs than the  
487 IPCC (2006) default value; however, other studies showed higher EFs (Table S1).  
488 These inconsistent reports may be reasonable, because a meta-analysis of global data  
489 found that riverine N<sub>2</sub>O EFs varied across regions (Hu et al. 2016). Meanwhile, our  
490 data showed significant negative relationships between EF and N loading (Figure 9),  
491 which is consistent with some previous studies (Beaulieu et al. 2011; Hinshaw and  
492 Dahlgren 2012; Hu et al. 2016). This can be attributed to decreasing microbial activity

493 with increasing N input due to progressive biological saturation (Hu et al. 2016;  
494 Mulholland et al. 2008). Although previous studies found that large rivers had low EF  
495 (Fu et al. 2018; Hinshaw and Dahlgren 2012), we proposed that the lower EF may  
496 appear in watersheds with a high anthropogenic N input (Figure 9), highlighting  
497 potential constraints in the IPCC methodology for rivers with high N loading. For  
498 example, in China, the annual mean riverine N<sub>2</sub>O EF in this agriculture watershed  
499 with a high N fertilizer rate (~600 kg N ha<sup>-1</sup> yr<sup>-1</sup>) was 0.0013, but in the center of the  
500 Sichuan Basin with a relatively low N fertilizer rate (~280 kg N ha<sup>-1</sup> yr<sup>-1</sup>) (Zhou et al.  
501 2012), the value was 0.0027 (Tian et al. 2017).

502

## 503 **5 Conclusions**

504 Two years' field measurements in the watershed, an intensively agricultural region  
505 located in Eastern China, showed the surface N<sub>2</sub>O concentrations and fluxes varied  
506 seasonally and spatially. The annual mean surface dissolved N<sub>2</sub>O concentrations for  
507 rivers, ponds, reservoir, and ditches were 30 nmol L<sup>-1</sup>, 19 nmol L<sup>-1</sup>, 16 nmol L<sup>-1</sup>, and  
508 58 nmol L<sup>-1</sup>, corresponding to the computed N<sub>2</sub>O fluxes of 49.6 μmol m<sup>-2</sup> d<sup>-1</sup>, 8.0  
509 μmol m<sup>-2</sup> d<sup>-1</sup>, 5.6 μmol m<sup>-2</sup> d<sup>-1</sup>, and 45.2 μmol m<sup>-2</sup> d<sup>-1</sup>, respectively.

510

511 Dissolved N<sub>2</sub>O concentration can be best predicted by NO<sub>3</sub><sup>-</sup>-N concentrations in  
512 rivers ( $r = 0.71, p < 0.02$ ) and by NH<sub>4</sub><sup>+</sup> in ponds ( $r = 0.65, p < 0.01$ ). The temporal  
513 variation of the riverine N<sub>2</sub>O concentration appears to be controlled by precipitation,  
514 and heavy precipitation induced the emission hot moments during the farming season.

515 Upstream waters are the hot spots in which riverine N<sub>2</sub>O production rates were two  
516 times greater than in non-hotspot locations.

517

518 The modeled watershed indirect N<sub>2</sub>O emission fluxes were comparable to the direct  
519 emission fluxes from the fertilized soils. A coarse estimate suggests that indirect N<sub>2</sub>O  
520 emissions represent approximately 4% of the total N<sub>2</sub>O emissions from N-fertilizer at  
521 the watershed scale. However, the IPCC default value of 0.0025 may overestimate the  
522 agriculture watershed indirect N<sub>2</sub>O emission in the study region. We propose that  
523 separate EFs for different seasons and different water types may be more appropriate  
524 due to the large variability.

525

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- 692

693 **Table 1.** Summary of the dissolved inorganic concentration ( $\text{NO}_3^-$ -N and  $\text{NH}_4^+$ -N),  
 694  $\text{N}_2\text{O}$  concentration and fluxes, and  $\text{N}_2\text{O}$  emission factor (EF) in all collected samples.  
 695 The presented values are the mean  $\pm$  standard deviation.

| Sample type | Season  | $\text{NH}_4^+$ -N<br>( $\text{mg L}^{-1}$ ) | $\text{NO}_3^-$ -N<br>( $\text{mg L}^{-1}$ ) | $\text{N}_2\text{O}$ concentration<br>( $\text{nmol L}^{-1}$ ) | $\text{N}_2\text{O}$ flux<br>( $\mu\text{mol m}^{-2} \text{d}^{-1}$ ) | EF                  |
|-------------|---------|--|--|--|---|---------------------|
| Rivers      | overall | 0.26 $\pm$ 0.15                              | 0.99 $\pm$ 0.59                              | 30 $\pm$ 18  | 49.6 $\pm$ 55.8   | 0.0013 $\pm$ 0.0010 |
|             | spring  | 0.16 $\pm$ 0.05                              | 0.49 $\pm$ 0.21                              | 19 $\pm$ 3   | 26.3 $\pm$ 12.4   | 0.0015 $\pm$ 0.0007 |
|             | summer  | 0.34 $\pm$ 0.15                              | 1.17 $\pm$ 0.94                              | 48 $\pm$ 28  | 107.9 $\pm$ 90.4  | 0.0021 $\pm$ 0.0016 |
|             | autumn  | 0.24 $\pm$ 0.13                              | 1.03 $\pm$ 0.38                              | 24 $\pm$ 8   | 31.3 $\pm$ 18.4   | 0.0009 $\pm$ 0.0005 |
|             | winter  | 0.30 $\pm$ 0.21                              | 1.26 $\pm$ 0.27                              | 28 $\pm$ 4   | 32.8 $\pm$ 13.0   | 0.0007 $\pm$ 0.0002 |
| Ponds       | overall | 0.32 $\pm$ 0.40                              | 0.47 $\pm$ 0.46                              | 19 $\pm$ 7   | 8.0 $\pm$ 8.6   | 0.0020 $\pm$ 0.0014 |
|             | spring  | 0.34 $\pm$ 0.43                              | 0.39 $\pm$ 0.33                              | 16 $\pm$ 3   | 5.8 $\pm$ 2.0   | 0.0015 $\pm$ 0.0006 |
|             | summer  | 0.37 $\pm$ 0.54                              | 0.20 $\pm$ 0.10                              | 16 $\pm$ 10  | 11.9 $\pm$ 15.9   | 0.0029 $\pm$ 0.0020 |
|             | autumn  | 0.25 $\pm$ 0.24                              | 0.50 $\pm$ 0.19                              | 20 $\pm$ 7   | 7.9 $\pm$ 5.9   | 0.0016 $\pm$ 0.0009 |
|             | winter  | 0.31 $\pm$ 0.39                              | 0.79 $\pm$ 0.73                              | 25 $\pm$ 5   | 6.0 $\pm$ 3.1   | 0.0016 $\pm$ 0.0011 |
| Reservoir   | overall | 0.19 $\pm$ 0.09                              | 0.54 $\pm$ 0.33                              | 16 $\pm$ 5   | 5.6 $\pm$ 5.5   | 0.0011 $\pm$ 0.0007 |
|             | spring  | 0.17 $\pm$ 0.04                              | 0.35 $\pm$ 0.15                              | 13 $\pm$ 3   | 2.8 $\pm$ 2.0   | 0.0012 $\pm$ 0.0007 |
|             | summer  | 0.24 $\pm$ 0.07                              | 0.58 $\pm$ 0.44                              | 18 $\pm$ 8   | 12.3 $\pm$ 6.9  | 0.0013 $\pm$ 0.0011 |
|             | autumn  | 0.18 $\pm$ 0.14                              | 0.56 $\pm$ 0.34                              | 14 $\pm$ 2   | 3.7 $\pm$ 2.1   | 0.0010 $\pm$ 0.0005 |
|             | winter  | 0.14 $\pm$ 0.07                              | 0.70 $\pm$ 0.29                              | 18 $\pm$ 2   | 3.0 $\pm$ 2.4   | 0.0008 $\pm$ 0.0004 |
| Ditches     | overall | 0.12 $\pm$ 0.15                              | 1.85 $\pm$ 1.81                              | 58 $\pm$ 69  | 45.2 $\pm$ 95.1   | 0.0014 $\pm$ 0.0013 |

696 **Figure 1.** Mean surface N<sub>2</sub>O concentration and calculated N<sub>2</sub>O flux for rivers, ponds,  
697 and reservoir during the two-year period. Error bars represent standard error. The  
698 symbols in the boxes indicate the hot moments for N<sub>2</sub>O, in which samples were  
699 collected from rivers after heavy precipitation during the farming season.

700

701 **Figure 2.** Temporal correlation between precipitation and mean concentrations of  
702 N<sub>2</sub>O and NO<sub>3</sub><sup>-</sup>-N in rivers. Note that the precipitation was the 10-day accumulated  
703 precipitation for each rainfall event before sampling. We assumed that major N losses  
704 from croplands induced by heavy rainfall might be transported to the rivers within 10  
705 days based on a previous study (Yan *et al.*, 2011)

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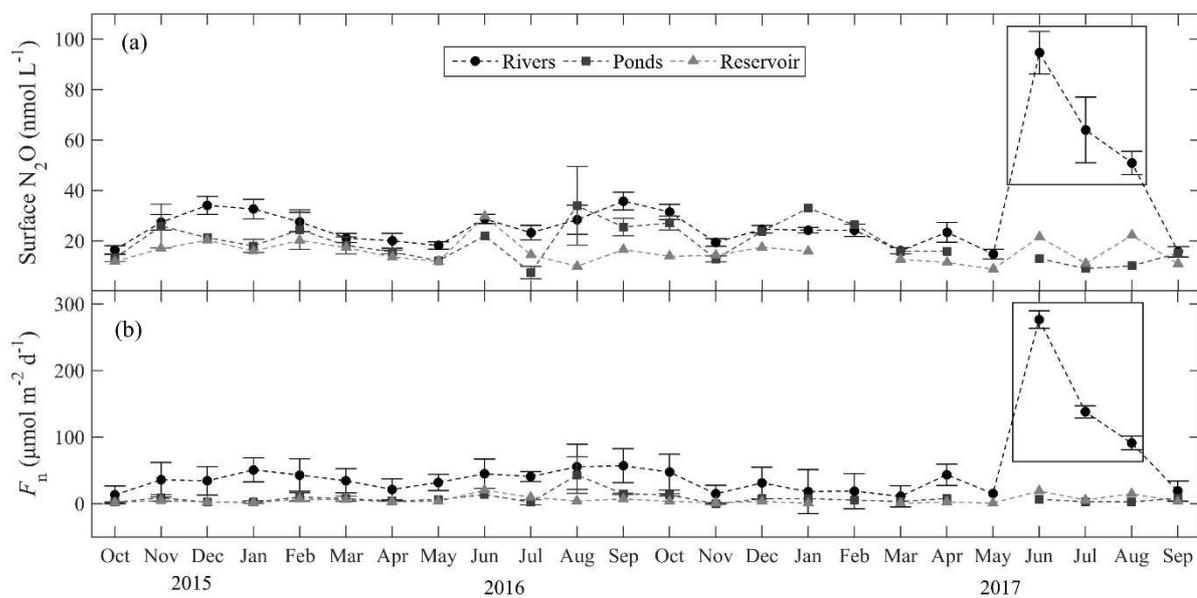
707 **Figure 3.** Temporal variability in the mean emission factor (N<sub>2</sub>O-N/NO<sub>3</sub><sup>-</sup>-N ratio) for  
708 rivers, ponds, and reservoir samples collected from November 2015 to September  
709 2017.

710

711 **Figure 4.** Simple liner regression of normalized monthly mean N<sub>2</sub>O emission factor  
712 (N<sub>2</sub>O-N/NO<sub>3</sub><sup>-</sup>-N ratio) and normalized NO<sub>3</sub><sup>-</sup>-N concentration in the collected rivers,  
713 ponds, and reservoir samples.

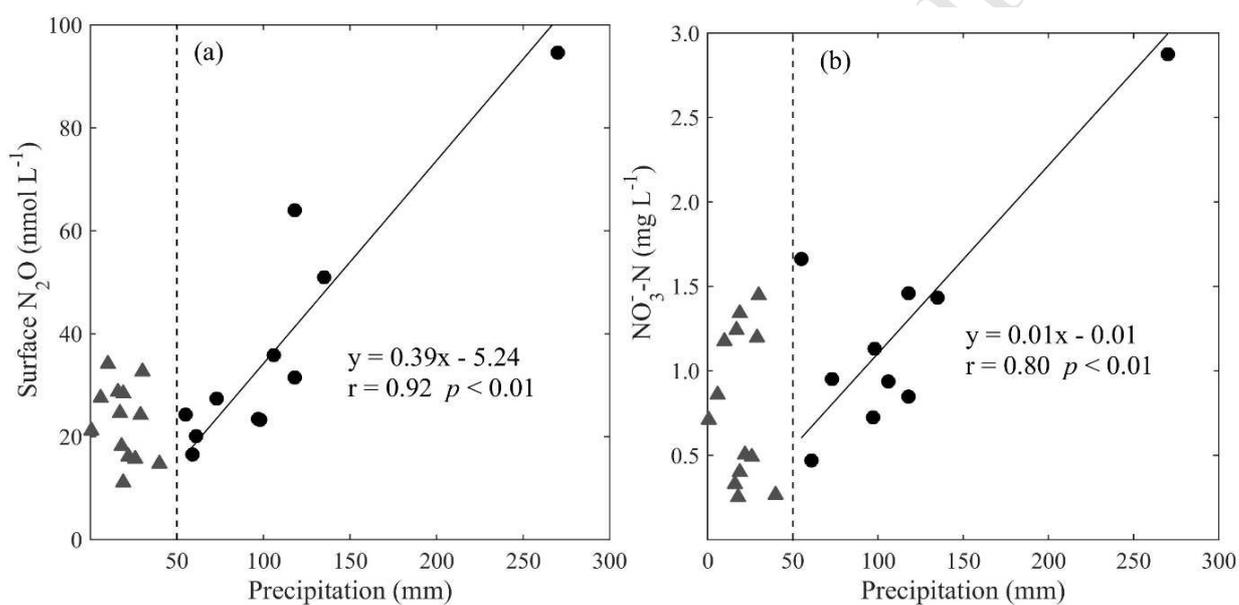
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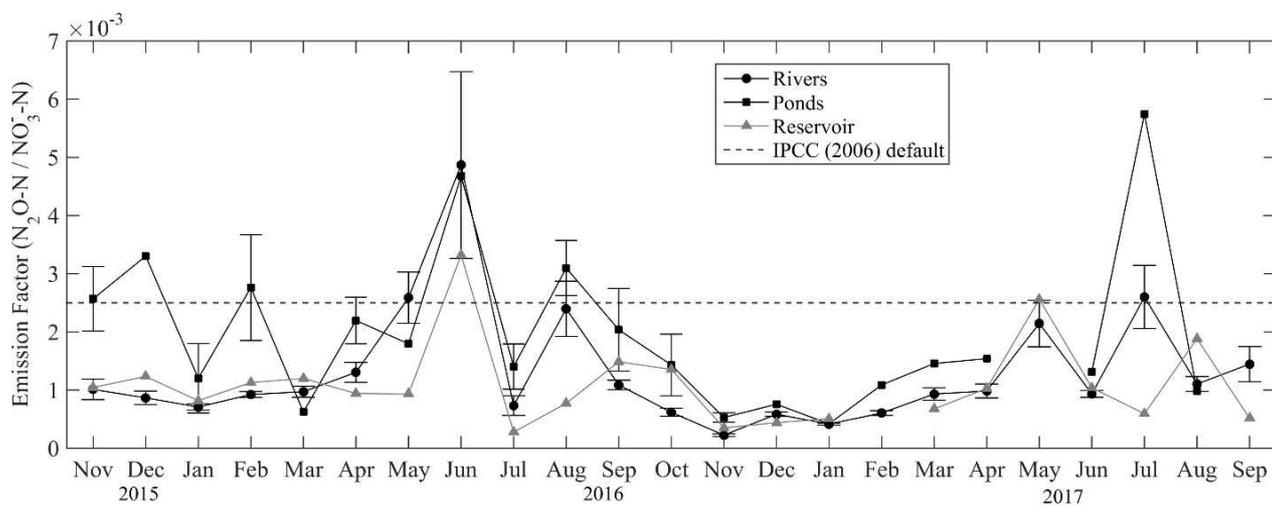
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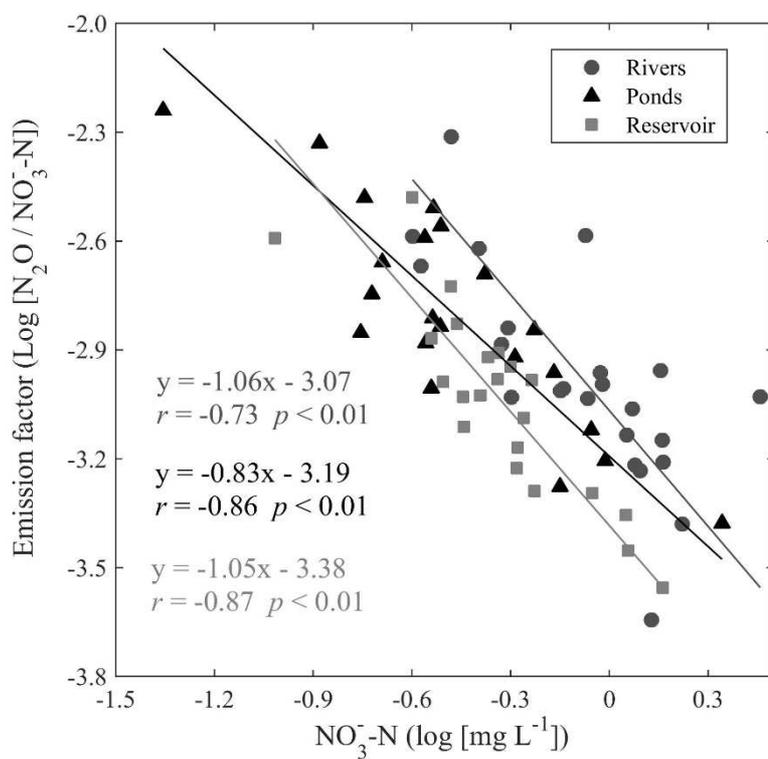
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728 **Figure 3.** Temporal variability in the mean emission factor ( $\text{N}_2\text{O-N}/\text{NO}_3^- \text{-N}$ ) for  
729 rivers, ponds, and reservoir samples collected from November 2015 to September  
730 2017.



733 **Figure 4.** Simple liner regression of normalized monthly mean N<sub>2</sub>O emission factor  
734 (N<sub>2</sub>O-N/NO<sub>3</sub><sup>-</sup>-N ratio) and normalized NO<sub>3</sub><sup>-</sup>-N concentration in the collected rivers,  
735 ponds, and reservoir samples.



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- N<sub>2</sub>O dynamics in water bodies of the agricultural watershed in eastern China, one intensive agricultural regions with heavy N fertilizer application, were investigated over a two-year period.
- Surface N<sub>2</sub>O concentrations and fluxes varied seasonally and spatially, which can be best predicted by N loadings and precipitation.
- The modeled watershed indirect N<sub>2</sub>O emission rates were comparable to direct emission from fertilized soil.
- IPCC method may overestimate the indirect N<sub>2</sub>O emission from agricultural watershed in eastern China.