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Surface nitrous oxide concentrations and fluxes from water bodies of the agricultural watershed in Eastern China^{\star}

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ABSTRACT

Agriculture is one of major emission sources of nitrous oxide (N₂O), an important greenhouse gas dominating stratospheric ozone destruction. However, indirect N₂O emissions from agriculture watershed water surfaces are poorly understood. Here, surface-dissolved N₂O concentration in water bodies of the agricultural watershed in Eastern China, one of the most intensive agricultural regions, was measured over a two-year period. Results showed that the dissolved N₂O concentrations varied in samples taken from different water types, and the annual mean N₂O concentrations for rivers, ponds, reservoir, and ditches were 30 ± 18 , 19 ± 7 , 16 ± 5 and 58 ± 69 nmol L⁻¹, respectively. The N₂O concentrations can be best predicted by the NO_3^- -N concentrations in rivers and by the NH_4^+ -N concentrations in ponds. Heavy precipitation induced hot moments of riverine N₂O emissions were observed during farming season. Upstream waters are hot spots, in which the N₂O production rates were two times greater than in nonhotspot locations. The modeled watershed indirect N₂O emission rates were comparable to direct emission from fertilized soil. A rough estimate suggests that indirect N₂O emissions yield approximately 4% of the total N₂O emissions yield from N-fertilizer at the watershed scale. Separate emission factors (EF) established for rivers, ponds, and reservoir were 0.0013, 0.0020, and 0.0012, respectively, indicating that the IPCC (Inter-governmental Panel on Climate Change) default value of 0.0025 may overestimate the indirect N_2O emission from surface water in eastern China. EF was inversely correlated with N loading, highlighting the potential constraints in the IPCC methodology for water with a high anthropogenic N input.

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1. Introduction

Nitrous oxide (N₂O) is a powerful and long-lived greenhouse gas and also a dominant component in the destruction of the stratospheric ozone (Ravishankara et al., 2009). The atmospheric N₂O concentration is estimated to have increased by 22% since 1750, with a current concentration of 329 ppb (Davidson, 2009; Cooper

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et al., 2017). The increasing atmospheric N₂O concentration has received considerable attention. The estimated annual global N₂O emission is 13.3–18.8 Tg N yr⁻¹, to which agriculture contributes nearly 30% (Xu et al., 2008; Cooper et al., 2017). Importantly, agriculture is considered by far the largest source (~80%) of anthropogenic N₂O via both direct emission and indirect emission (Kroeze et al., 1999; Davidson, 2009; Cooper et al., 2017).

Anthropogenic N₂O input into the atmosphere from fertilized soils are defined as direct emission, while indirect emission represents N₂O production in ditches, streams, and rivers induced by leaching and runoff of reactive N from agricultural areas (Mosier et al., 1998; Fu et al., 2018). Direct N₂O emissions have been well documented through a large amount of field measurements (Xu et al., 2008; Shcherbak et al., 2014), while indirect emissions from







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agricultural watershed are less well constrained due to the scarcity of studies on this topic (Reay et al., 2003; Outram and Hiscock; Fu et al., 2018). Importantly, indirect emission is estimated to account for over one-quarter of global total agricultural N₂O emissions (Reay et al. 2012; Venkiteswaran et al., 2014) and dominates the inter-annual variability of total emission (Griffis et al. 2017). Meanwhile, riverine N₂O emissions, especially in agricultural areas, remain a major source of uncertainty in the global N₂O budget due to the large spatiotemporal variability (Beaulieu et al. 2011; Borges et al. 2015; Audet et al., 2017). More field measurements of indirect N₂O emission are needed to reduce the uncertainty and improve the reliability of the global anthropogenic N₂O budget (Reay et al. 2012; Saikawa et al. 2014).

IPCC provided emission factors (EF) to estimate indirect N₂O emissions resulting from fertilizer-N and manure-N loss via leaching and runoff (De Klein et al. 2006), in which global and regional N₂O emissions from a water body can be calculated by multiplying the nitrogen fertilizer input or relevant anthropogenic N loading with the default EF (Hu et al., 2016; Hama-Aziz et al., 2017; Fu et al., 2018). Since 2006, the IPCC default EF values for rivers, groundwater, and estuaries were both set to 0.0025 (De Klein et al., 2006). However, studies have shown that the default EF is poorly constrained due to high variability in environmental conditions (Turner et al., 2015; Cooper et al., 2017; Griffis et al. 2017). For example, a field measurement showed that the IPCC default EF is underestimated up to nine-fold in rivers in the U.S. Corn Belt, an intensively agricultural region (Turner et al., 2015; Fu et al., 2017; Fu et al., 2018), but it is unexpectedly overestimated in an intensively agriculture area in other regions (Clough et al., 2007: Outram and Hiscock, 2012; Hama-Aziz et al., 2017). Using the default IPCC emission factor for predicting indirect N₂O emissions from all rivers may be inappropriate, and more measurements are needed to accurately calculate the EF_{5r} (Outram and Hiscock, 2012; Hama-Aziz et al., 2017; Tian et al., 2017).

Heavy farmland N fertilizer application in Eastern China, one of most intensively agricultural regions in the world, has led to a high N loading input in the watershed. The fertilizer N application rate is up to 600 kg N ha⁻¹ yr⁻¹ in Eastern China (Ju et al. 2009; Xing and Zhu, 2002). For comparison, the value is only approximately 100 kg N ha⁻¹ yr⁻¹ in intensively agricultural regions in the US (Griffis et al. 2017; Turner et al., 2015). Further, a large amount of farmland N-fertilizer (~280 kg N ha⁻¹ yr⁻¹) enters the watershed via runoff and leaching in Eastern China (Xing and Zhu, 2002; Yan et al. 2011), which may significantly increase the N₂O production rates (Beaulieu et al. 2011). Previous studies suggested that the surface water N₂O emissions in Eastern China may be very high due to the vast farmland N-fertilizer (Seitzinger and Kroeze, 1998). However, until now, limited information is available regarding the characteristics of watershed N₂O dynamics in this area.

In this study, we measured the dissolved N₂O concentration in water bodies in a typical agricultural watershed in Eastern China. Our objectives were to quantify the importance of agricultural watershed fueled by farmland N-fertilizer as sources of indirect N₂O emission in this area, to explore the indirect N₂O emission factor, and to further investigate the spatiotemporal characteristics of N₂O dynamics.

2. Materials and methods

2.1. Study area

This field measurement was carried out in a typical agriculture watershed located in Eastern China (31°58' to 32°01'N, 119°12' to 119°14'E; Fig. S1). The watershed has previously been reported in detail (Yan et al. 2011). Briefly, there is a reservoir, three rivers, and

thousands of small pond in the watershed. The average annual precipitation is 1100 mm and the mean annual temperature is 15 °C. The major annual cropping rotations are rice-wheat for the paddy fields and maize-rapeseed, respectively. There is no industry in the studied area, and agriculture is the dominant local source of anthropogenic N.

2.2. Sample collection and analysis

A watershed-scale sampling round was conducted in each month from October 2015 to September 2017, during which samples were taken from the surface water of the reservoir, rivers, ponds, and ditches (Fig. S1). As a whole, water samples were collected from 30 sampling sites, of which 1 site was in a reservoir, 19 sites were in rivers, 3 sites were in ponds, and 7 sites were in ditches. There are three rivers in the watershed: River 1, River 2, and River 3, respectively. Given the river length, the sampling sites were divided into two sections in River 1 (midstream and downstream) and three sections (upstream, midstream, and downstream) in River 2, respectively. It should be noted that the ditches sometimes dry up without covering water, leading to discontinuous sampling.

Each watershed-scale survey was completed over two consecutive days. The procedures of sampling and analysis have previously been reported in detail in Xiao et al., (2019). Briefly, triplicate bubble-free water samples were collected at a 20-cm depth below the surface using 300 mL glass bottles, and these bottles were transported to lab for dissolved N₂O concentration analysis using the headspace equilibration method. Water samples were also collected using an organic glass hydrophore to the ammonium-nitrogen (NH_4^+ -N), nitrate-nitrogen (NO_3^- -N), and nitrite-nitrogen (NO_2^- -N) analysis by a continuous flow analyzer (Skalar san⁺⁺, Netherlands). The water temperature, pH, specific conductance (Spc), and dissolved oxygen concentration (DO) were measured *in situ* with a multiparameter probe (YSI 650MDS, YSI Inc. Yellow Springs, OH, USA) at a 20-cm depth to be consistent with water sampling.

2.3. N₂O fluxes and emission factor calculations

The N₂O fluxes (F_n , µmol m⁻² d⁻¹, positive indications of N₂O emission from water to atmosphere) was calculated using water-air gas exchange model:

$$F_{\rm n} = k \times (C_{\rm w} - C_{\rm eq}) \tag{1}$$

where C_w is the surface dissolved N₂O concentration (µmol m⁻³) in water measured by headspace equilibration method, C_{eq} is the N₂O concentration in water that is in equilibrium with the atmosphere at the *in-situ* temperature, and *k* is the gas transfer coefficient. For the rivers, *k* was calculated considering both wind and water turbulence (Clough et al., 2007):

$$k = 2.78E^{-6}au_{10}^2 \left(\frac{S_c}{660}\right)^{0.5} + \sqrt{\frac{DU}{h}}$$
(2)

where $2.78E^{-6}$ is a conversion factor (cm h⁻¹ to m s⁻¹), *a* is a constant (0.31), u_{10} is the wind speed at a height of 10 m above the water surface, and S_c is the Schmidt number for N₂O, *U* is the river water velocity (m s⁻¹), *h* is the average river depth (m), and *D* is the N₂O diffusion coefficient in the water. The hourly measurement of wind speed was obtained from a weather station in the watershed to calculate the *k*. The river water velocity is steady and only varies depending upon precipitation (Yan et al. 2011; Xia et al., 2013b).

For the lentic ecosystem (reservoir and ponds), the N₂O transfer

velocity is mainly driven by wind speed because no surface water flow occurs. k (cm h⁻¹) was calculated from a wind-dependent formula derived from a small shallow lake as follows (Cole and Caraco, 1998):

$$k = (S_{\rm c}/660)^{-n} \left(2.07 + 0.215 \times u_{10}^{1.7}\right) \tag{3}$$

There is no separate equation for k calculation for ditches, and Equation (3) was also used for calculating the gas transfer coefficient k for ditches.

The indirect N_2O emission factor (EF) for rivers, ponds, ditches, and the reservoir were calculated using the common IPCC methodology as follows (De Klein et al. 2006):

$$EF = N_2 O - N NO_3^- - N$$
 (4)

Where N₂O-N and NO_3^- -N are concentrations measured in rivers, ponds, ditches, and the reservoir. Equation (4) has been widely used to calculate the indirect N₂O emission factor (Turner et al., 2015; Hama-Aziz et al., 2017; Tian et al., 2017).

2.4. Data analysis

Statistical analyses were conducted with SPSS version 18.0 (SPSS, Inc., USA).

In the study area, spring is from March to the end of May, summer is from June to August, autumn is from September to November, and winter is from December to February in next year. Simple linear and multi-linear stepwise regression analyses were performed to determine the relationships among N₂O, the emission factor, and independent variables (e.g., NH_4^+ -N, NO_3^- -N, and NO_2^- -N). The entire two-year dataset was divided according to water type (e.g., rivers, ponds, ditches, and reservoir), and the difference of their mean N₂O concentrations and fluxes was examined using a one-way analysis of variance (ANOVA). After completing the ANOVA test, a least significant difference (LSD) post hoc test was conducted. Differences at *p* < 0.05 were labeled as statistically significant in all analyses.

3. Results

3.1. Physical and chemical characteristics

There was no obvious variation in water temperature across

different water body types in the agricultural watershed (Fig. 2S(a)). The annual mean water temperature is 19.3 °C over the two-year study period. There was a clear seasonality for water temperature, with the highest water temperature appearing in the summer (33.3 °C) and the lowest (6.3 °C) in winter. The annual mean wind speed was 2.5 m s^{-1} based on the measurement from the local weather station.

In contrast to water temperature, the surface water chemical properties varied among rivers, ponds, reservoir, and ditches (Table 1 and Fig. 2S). Generally, the NO_3^- -N concentrations was highest in ditches with annual mean value of 1.85 mg L^{-1} , followed by rivers (0.99 mg L^{-1}), reservoir (0.54 mg L^{-1}) and ponds (0.47 mg L^{-1}). In contrast, the highest NH_4^+ -N concentration appeared in ponds with annual mean value of 0.32 mg L^{-1} , followed by rivers and reservoirs. The NO_2^- -N concentrations in these surface water areas showed minor differences. The average DO concentration was over-saturated only in the reservoir with a value of 10.45 mg L⁻¹ or 110% saturation, and the annual mean values for rivers, ponds, and ditches were 6.65, 5.78, and 7.90 mg L⁻¹, respectively.

The NO_3^- -N was the dominant form of the total dissolved inorganic nitrogen (DIN) in the surface water of the agricultural watershed (Table 1). For example, the NO_3^- -N concentration accounted for 76%, 57%, 72%, and 92% of the total DIN concentration for rivers, ponds, the reservoir, and ditches, respectively. In addition, the surface water NO_3^- -N concentrations for the rivers, ponds, and reservoir showed clear seasonality with peak concentrations being observed in summer and winter (Fig. 2S(b)), coincides with the crop-planting period. There were no obvious temporal variations for NH_4^+ -N concentration except in ponds.

3.2. Spatial and temporal characteristics of surface water N_2O concentration

The dissolved N₂O concentration in surface water varied across water body types (Fig. S3). Dissolved N₂O concentrations in the ditches spanned a broad range, from 10 to 243 nmol L⁻¹, but varied within relatively narrow ranges of 4–172 nmol L⁻¹, 5–50 nmol L⁻¹, and 9–30 nmol L⁻¹ in rivers, pond, and reservoir, respectively. Meanwhile, the annual mean surface N₂O concentration in ditches with a value of 63 nmol L⁻¹ was significantly (p < 0.01) higher than those in rivers (29 nmol L⁻¹), ponds (20 nmol L⁻¹), and reservoir (15 nmol L⁻¹). The differences of N₂O concentrations between



Fig. 1. Mean surface N₂O concentration and calculated N₂O flux for rivers, ponds, and reservoir during the two-year period. Error bars represent standard error. The symbols in the boxes indicate the hot moments for N₂O, in which samples were collected from rivers after heavy precipitation during the farming season.



Fig. 2. Temporal correlation between precipitation and mean concentrations of N_2O and NO_3^- -N in rivers. Note that the precipitation was the 10-day accumulated precipitation for each rainfall event before sampling. We assumed that major N losses from croplands induced by heavy rainfall might be transported to the rivers within 10 days based on a previous study (Yan *et al.*, 2011).



Fig. 3. Temporal variability in the mean emission factor (N₂O-N/NO₃⁻-N ratio) for rivers, ponds, and reservoir samples collected from November 2015 to September 2017.

Table 1

Summary of the dissolved inorganic concentration (NO_3^- -N and NH_4^+ -N), N₂O concentration and fluxes, and N₂O emission factor (EF) in all collected samples. The presented values are the mean \pm standard deviation.

Sample type	Season	NH_4^+ -N (mg L ⁻¹)	NO_3^- -N (mg L ⁻¹)	N_2O concentration (nmol L ⁻¹)	N_2O flux (µmol m ⁻² d ⁻¹)	EF
Rivers	overall	0.26 ± 0.15	0.99 ± 0.59	30 ± 18	49.6 ± 55.8	0.0013 ± 0.0010
	spring	0.16 ± 0.05	0.49 ± 0.21	19±3	26.3 ± 12.4	0.0015 ± 0.0007
	summer	0.34 ± 0.15	1.17 ± 0.94	48 ± 28	107.9 ± 90.4	0.0021 ± 0.0016
	autumn	0.24 ± 0.13	1.03 ± 0.38	24 ± 8	31.3 ± 18.4	0.0009 ± 0.0005
	winter	0.30 ± 0.21	1.26 ± 0.27	28 ± 4	32.8 ± 13.0	0.0007 ± 0.0002
Ponds	overall	0.32 ± 0.40	0.47 ± 0.46	19 ± 7	8.0 ± 8.6	0.0020 ± 0.0014
	spring	0.34 ± 0.43	0.39 ± 0.33	16 ± 3	5.8 ± 2.0	0.0015 ± 0.0006
	summer	0.37 ± 0.54	0.20 ± 0.10	16 ± 10	11.9 ± 15.9	0.0029 ± 0.0020
	autumn	0.25 ± 0.24	0.50 ± 0.19	20 ± 7	7.9 ± 5.9	0.0016 ± 0.0009
	winter	0.31 ± 0.39	0.79 ± 0.73	25 ± 5	6.0 ± 3.1	0.0016 ± 0.0011
Reservoir	overall	0.19 ± 0.09	0.54 ± 0.33	16 ± 5	5.6 ± 5.5	0.0011 ± 0.0007
	spring	0.17 ± 0.04	0.35 ± 0.15	13 ± 3	2.8 ± 2.0	0.0012 ± 0.0007
	summer	0.24 ± 0.07	0.58 ± 0.44	18 ± 8	12.3 ± 6.9	0.0013 ± 0.0011
	autumn	0.18 ± 0.14	0.56 ± 0.34	14 ± 2	3.7 ± 2.1	0.0010 ± 0.0005
	winter	0.14 ± 0.07	0.70 ± 0.29	18 ± 2	3.0 ± 2.4	0.0008 ± 0.0004
Ditches	overall	0.12 ± 0.15	1.85 ± 1.81	58 ± 69	45.2 ± 95.1	0.0014 ± 0.0013

rivers, ponds, and reservoir were insignificant (p > 0.05).

The N₂O concentrations also varied within rivers. River 3 with the shortest transport length (Fig. S1) had the highest N₂O concentration compared to River 1 and River 2. The overall mean N₂O concentrations were 26, 30, and 35 for rivers 1, 2, and 3, respectively. Further analysis showed large spatial variations for the riverine N₂O concentration along the current direction of the water (Fig. S4). A N₂O concentration gradient from the upstream to the estuary is clearly evident. Peak N₂O concentrations were typically found in the upstream of the river. The annual mean N₂O concentrations in upstream, midstream, and downstream of River 2 were 45, 25, and 21 nmol L^{-1} , respectively. River 3 is too short to be

sampled at different locations, and samples were only taken from midstream and downstream in River 1, but the measured data also showed that the N₂O concentrations decreased rapidly far from the upstream areas (Fig. 4S(b)).

Another notable feature was that the surface N₂O concentration varied over time, especially in rivers. Hot moments, or rapid temporal increases in the surface N₂O concentrations, were evident in rivers (Fig. 1). There was clear seasonality for the monthly mean N_2O concentration in rivers, ranging from 15 to 95 nmol L⁻¹, with peak concentrations being observed in the summer. In contrast to rivers, the temporal variation showed a narrower range for ponds $(7-34 \text{ nmol } L^{-1})$ and the reservoir $(9-30 \text{ nmol } L^{-1})$ but with peak concentrations being observed mostly in the winter. For rivers, the summertime N_2O concentrations with a mean value of 48 nmol L^{-1} were significantly (p < 0.05) higher than those in spring $(19 \text{ nmol } L^{-1})$, autumn $(24 \text{ nmol } L^{-1})$, and winter $(28 \text{ nmol } L^{-1})$. For ponds, the summertime N₂O concentration with a mean value of 16 nmol L⁻¹ was significant (p < 0.05) lower than that in winter (25 nmol L^{-1}) . Additionally, the seasonal average values in the reservoir were 13 (spring), 18 (summer), 14 (autumn), and 18 nmol L⁻¹ (winter), respectively, showing significant (p = 0.046) difference between spring and summer. We cannot characterize the seasonal variation of N2O concentrations in ditches due to noncontinuous measurements in time.

3.3. Relationships between N_2O concentration and environment factors

Surface N₂O concentration in the rivers, ponds, and reservoir exhibited different responses to inorganic N loadings (Fig. S5). The N₂O concentration and NO_3^- -N concentration exhibited a stronger correlation in rivers (r = 0.71, p < 0.01) than in ponds (r = 0.38, p = 0.03). Inversely, a stronger correlation between N₂O concentration and NH_4^+ -N concentration was found in ponds (r = 0.65, p < 0.01) than in rivers (r = 0.33, p < 0.01). Additionally, the NO_2^- -N concentration was well correlated with the riverine N₂O concentration (r = 0.65, p < 0.01). In the reservoir, the N₂O concentrations showed no significant correlation with any form of inorganic N loadings.



Fig. 4. Simple liner regression of normalized monthly mean N_2O emission factor (N_2O - N/NO_3^- -N ratio) and normalized NO_3^- -N concentration in the collected rivers, ponds, and reservoir samples.

The temporal correlation between the surface water N₂O concentration and water temperature were insignificant for rivers (r = 0.27, p = 0.20), ponds (r = 0.37, p = 0.08), and reservoir (r = 0.11, p = 0.08)p = 0.61). Correlation analyses also revealed that the temporal variation in N₂O concentration was not correlated well with the water chemical properties (e.g., Spc, pH, and DO). Precipitation, by contrast, appeared to contribute to the temporal variation of the N₂O concentration (Fig. 2). The accumulated precipitation for ten days before sampling was well correlated with the monthly riverine N₂O concentration when the precipitation amounts were greater than 50 mm (r = 0.92, p < 0.01). Importantly, the hot moments, the rapid temporal increase in the riverine N₂O production (Fu et al., 2018), were induced by heavy precipitation during the cropplanting period in the study (Fig. 1). However, the temporal correlation between precipitation and the N₂O concentration in the lentic ecosystems (ponds and reservoir) was insignificant (p > 0.05).

3.4. Indirect N₂O fluxes

The mean estimated indirect N₂O emission (mean \pm 1 standard deviation) from the surface water of the agricultural watershed was 27.1 \pm 23.4 µmol m⁻² d⁻¹ based on two years' field measurements. The rivers were found to have the highest N₂O fluxes, with a mean value of 49.6 µmol m⁻² d⁻¹. Ditches had the second highest N₂O fluxes with large uncertainly (mean \pm 1 standard deviation: 45.2 \pm 95.1 µmol m⁻² d⁻¹; Table 1). The ponds and reservoir showed moderate emission fluxes with a mean value of 8.0 and 5.6 µmol m⁻² d⁻¹, respectively. The highest N₂O fluxes for the three water body types — rivers, ponds, and reservoir — were found in the summer (Table 1).

The indirect N₂O fluxes from water bodies in the watershed were calculated with the dissolved N₂O concentration. The calculated indirect N₂O fluxes and dissolved N₂O concentration were highly correlated for all records across water body types and over time (r = 0.97, p < 0.01). The roles of environment factors influencing the N₂O fluxes were similar to those of surface water-dissolved N₂O concentration.

3.5. Indirect N₂O emission factor

The results revealed that the EF was not uniform for the different water types (Fig. 3). Ponds had the highest EF, followed by ditches and rivers on average (Table 1). The reservoir, the large water body in the watershed, had the lowest EF with a mean value of 0.0011. The EF of 0.0013 for rivers was nearly half of the default 0.0025 by IPCC. Additionally, the measured EF of 0.0014 for ditches was also significantly lower than the IPCC default value of 0.0025 for river.

It was evidenced from the field measurements that EF varied over time (Fig. 3). The monthly mean EF varied from 0.0002 to 0.0049 for rivers, 0.0004 to 0.0057 for ponds, and 0.0002 to 0.0033 for the reservoir over the two-year period. The riverine EF showed apparent seasonality: summer (mean ± 1 standard deviation: 0.0021 ± 0.0016) > spring (0.0015 ± 0.0007) > autumn (0.0009 ± 0.0005) > winter (0.0007 ± 0.0002 , Table 1). In contrast to N₂O concentration, the EFs were highly negative with the NO_3^- -N concentration for rivers, ponds, and the reservoir (Fig. 4).

4. Discussion

4.1. Controls on N₂O variability

Previous field measurements demonstrated that although other environmental factors such as dissolved oxygen (Kampschreur et al., 2009; Rosamond et al., 2012), organic carbon (Kampschreur et al., 2009), and water temperature (Venkiteswaran et al., 2014) affect the riverine N₂O production rate, the NO₃⁻-N concentration was the predominant control on the riverine N₂O concentration in this study (Fig. S5(a)). The pattern was consistent with previous observations, showing that NO₃-N is an important driver of riverine N₂O production in agricultural watersheds (Beaulieu et al., 2008: Baulch et al., 2011: Turner et al. 2016: Hama-Aziz et al., 2017: Tian et al., 2017). It is likely that farmland N fertilizer loading to rivers stimulates denitrification and associated N₂O production (Kampschreur et al., 2009; Beaulieu et al. 2011; Maavara et al., 2019). NH_{4}^{+} -N concentrations, a good indicator of domestic pollution (Garnier et al. 2009), were found to peak in ponds (Table 1). Additionally, the strong relationship between NH_4^+ -N and pond N₂O concentration (r = 0.65, p < 0.01; Fig. S5(b)) is reasonable because the high NH_4^+ -N level may increase the N₂O accumulation via nitrification, which had been demonstrated in high population density areas (Xia et al., 2013a; Yu et al. 2013; Hu et al., 2016). The poor correlation between N₂O and any form of N loading in the reservoir, by contrast, indicated that other biotic and abiotic factors played a greater role than N loading in N₂O production (Deemer et al. 2016).

Both field measurement and model simulation find that the riverine N₂O production rate is highly sensitive to perturbations in precipitation (Chen et al. 2016; Cooper et al., 2017; Griffis et al. 2017; Fu et al., 2018). Here, precipitation (10-day accumulated precipitation prior to sampling > 50 mm) was found to drive the temporal variation of the riverine N₂O concentration (Fig. 2a), consistent with previous studies in the agriculture watersheds in Southwestern China (Tian et al., 2017) and UK (Cooper et al., 2017). Heavy precipitation is likely to transport more agricultural N loading to the river (Sinha et al., 2017, Fig. 2b) and subsequently stimulates N₂O production (Liu and Greaver, 2009; Tian et al., 2017). The results reported here showed that heavy precipitation combined with intense agricultural fertilizer application during the farming season triggered riverine N₂O hot moments (Fig. 1), which should be captured to reduce the uncertainty of the N₂O budget (Fu et al., 2018; Shrestha and Wang, 2018). The projected increase in both heavy and total precipitation induced by climate change (Sinha et al., 2017), together with the increase in the use of nitrogen fertilizer to meet the food demand (Jiang et al., 2010; Saikawa et al. 2014), posed significant potential for agricultural watershed indirect N₂O production in the global N₂O budget.

Although many previous studies showed a strong positive correlation between temperature and riverine N₂O production (Beaulieu et al., 2010; Hinshaw and Dahlgren, 2012; Shrestha and Wang, 2018), our results reported here were consistent with the field measurement in a Swedish agricultural catchment showing water temperature had an insignificant effect on seasonal variation of N₂O concentration (Audet et al., 2017). This is probably due to the N concentration outweighs the temperature in determining the water N₂O production via nitrification and denitrification (Kampschreur et al., 2009; Davidson et al. 2015; Capodici et al., 2018). Another notable feature was that the surface NO_3^- -N concentration in rivers was the highest in winter, but the highest N₂O concentration occurred in summer in this study (Table 1). Some studies have proposed that the carbon limitation during denitrification is associated with increased N₂O production (Kampschreur et al., 2009; Capodici et al., 2018), which may be the explanation for this pattern. Because study has shown that the dissolved organic carbon in these rivers increased from 6.2 mg L^{-1} in summer to 18.6 mg L^{-1} in winter, with the C:N ratio increased from 1.3 to 3.1 (Zhao et al., 2013). The high C:N ratio may supply availability carbon sources, resulting in an increase in denitrification efficiency and associated decrease in N₂O production although the high NO₃⁻-N concentration in winter in this study (Kampschreur et al., 2009;

Zhao et al., 2013; Capodici et al., 2018).

The riverine N₂O production rate varies greatly by geographic location (Fig. S4). The most notable feature was that the N₂O concentration decreased sharply from the upstream to the estuary, with the mean value decreasing from 47 to 17 nmol L^{-1} in River 2. The result reported here is consistent with the field measurement in other intensive agricultural activity areas, such as the Seine basin in France (Garnier et al. 2009), the U.S. Corn Belt (Turner et al., 2015), and a typical agriculture watershed in the UK (Reay et al., 2003). The high NO_3^- -N concentrations in upstream waters (Fig. 4S(a)), may enhance the microbial processes of N₂O production (Herrman et al., 2008; Zhao et al., 2014). Notably, direct N₂O input from other sources, such as soil and groundwater, might also lead to a high N₂O concentration in upstream waters (Garnier et al. 2009; Beaulieu et al. 2011). As suggested in previous studies, the upstream waters are the hot spot for N₂O production (Peterson et al. 2001; Turner et al., 2015). In summary, N₂O production rates in hotspots were two times greater than in non-hotspot locations in the presented study.

4.2. Indirect N₂O fluxes in the agricultural watershed

The combined indirect N₂O fluxes from rivers, ponds, and reservoir were calculated to be 383 kg N₂O-N yr⁻¹. The rivers were the most significant contributor of N₂O fluxes of all water bodies. contributing to a total of 161 kg N₂O-N yr⁻¹. Total direct N₂O fluxes from the watershed fertilized soil was estimated to be $12 t (10^3 kg)$ N vr^{-1} in the watershed (Xia et al., 2013b). Based on a study in another similar region in Eastern China (She et al. 2018), we roughly estimated that the ditch area for the rice-wheat field is 25 hm². Adding the annul N₂O emission yield from agriculture ditches $(\sim 117 \text{ kg N}_2\text{O}-\text{N yr}^{-1})$ in the watershed, the watershed indirect N₂O emission thus represents approximately 4% of the total N₂O emissions from N-fertilizer. Although the obtained percentage was similar to other studies (Beaulieu et al., 2008; Garnier et al. 2009; Audet et al., 2017), caution should be taken for the following reasons: (1) the ditch areas and N₂O emission rates $(45.2 \pm 95.1 \,\mu\text{mol}\,\text{m}^{-2}\,\text{d}^{-1})$ had large uncertainly; (2) the presented sampling strategies might omit some riverine emission hot spots and hot moments, which may lead to low biases in the annual total N₂O emission yield (Fu et al., 2018).

For lentic ecosystems, the annual mean N₂O emission rates in ponds and reservoir (8.0 and 5.6 μ mol m⁻² d⁻¹, respectively) were higher than the median emission rate (3.9 μ mol m⁻² d⁻¹) in global lakes/reservoirs according to Hu et al. (2016). Meanwhile, these values were also higher than those in temperature lakes with a mean value of 3.4 μ mol m⁻² d⁻¹ (Soued et al., 2016). A heavily polluted lake nearby, Lake Taihu, showed a moderate N₂O emission of 3.5 μ mol m⁻² d⁻¹ (Xiao et al., 2019). The gas transfer velocity of a lentic ecosystem in the study region can be accurately estimated using the wind-dependent formula, which had been demonstrated previously (Xiao et al. 2017). These results together indicated watershed N inputs fueled the pond and reservoir N₂O emission.

The indirect riverine N₂O fluxes reported here is compared to the results reported by previous studies (Table S1). The literature review in Table S1 showed that the riverine N₂O fluxes across multiple land-use types ranged from 6.7 to 121.3 μ mol m⁻² d⁻¹, with a mean value of 51.3 μ mol m⁻² d⁻¹. Based on a meta-analysis with 169 observations, the global median riverine N₂O emission fluxes is 14.4 μ mol m⁻² d⁻¹, which is significantly lower than the annual mean flux reported here (49.6 μ mol m⁻² d⁻¹). For comparison, the direct N₂O emission rate from fertilized soil in the study area is approximately 60–90 μ mol m⁻² d⁻¹ (Zou et al., 2005; Zhou et al. 2014; Liu et al., 2016). Therefore, indirect N₂O emissions caused by N leaching and surface runoff N losses in agriculture watershed could contribute significantly to the total $N_2 O\ emission.$

The N₂O emissions from water are often determined by waterair gas exchange model or direct field measurement such as using floating chamber technique (Outram and Hiscock, 2012; Zhu et al. 2015; Hama-Aziz et al., 2017; Wu et al., 2018; Capodici et al., 2018). The water-air gas exchange model has been frequently employed to quality N₂O emission flux from water, accounting for 80% of flux measurements (Hu et al., 2016; Wu et al., 2018), however, the model-based method may either overestimate or underestimate N₂O emission due to the uncertainly in gas transfer coefficient (Yu et al. 2013; Raymond et al. 2013; Wu et al., 2018). For comparison, the direct measurement using floating chamber method showed the N₂O emission flux were $48.1-66.1 \,\mu\text{mol}\,\text{m}^{-2}$ d^{-1} and below 8.6 μ mol m⁻² d^{-1} from the rivers and ponds, respectively, within the same watershed (Li et al., 2011; Xia et al., 2013a; Han et al., 2014), which were similar to the modelestimated N₂O emission flux in the presented study (river: 49.6 μ mol m⁻² d⁻¹; ponds: 8.0 μ mol m⁻² d⁻¹). Additionally, previous study has shown that the N₂O emission flux measured by the two methods above exhibits the same seasonal variation in an agricultural watershed in southeast China (Wu et al., 2018). Both of these indicate the model-based method was reasonable for guantifying the watershed indirect N₂O emission in this study.

4.3. Implication of the measured EF values

This study revealed that indirect N_2O EFs were not uniform for the different surface water types within the same watershed (Table 1 and Fig. 3). This is consistent with the field measurement in UK agriculture watershed, which shows that different water types produce various amounts of N_2O (Outram and Hiscock, 2012; Hama-Aziz et al., 2017). Therefore, separate EFs for each water compartment are required to better understand the indirect N_2O budget. Emission factor for lentic ponds and the reservoir should also be included because our results showed that ignoring the indirect N_2O emission from the lentic ecosystems would lead to an underestimation of the indirect emission yield by 60%.

For rivers and ditches, the EF values of 0.0013 and 0.0014 were both lower than the IPCC EF value of 0.0025. These results indicated that the IPCC default value of 0.0025 may overestimate the agriculture watershed indirect N₂O emission in the study region. The recent study by Maavara et al., (2019) using mechanistic modeling approach also showed IPCC EFs are likely overestimated. Meanwhile, our data showed that the EF value across different water types varied over time with a peak appearing in the summer (Fig. 3), which is consistent with a previous study (Hama-Aziz et al., 2017). Thus, it may be more applicable to have more measurements across different seasons. Meanwhile, we also found that the IPCC approach of using one EF for all rivers may be inappropriate because EFs were highly variable across rivers, as suggested by Beaulieu et al. (2008). The EF values for rivers 1 2, and 3 were 0.0015, 0.0011, and 0.0016, respectively. Since there are no IPCC EFs for the ponds and reservoir (Outram and Hiscock, 2012), a comparison cannot be made. However, a synthesis analysis showed that EF for ponds and reservoirs was 0.0012 (Tian et al., 2018), which was significantly lower than that reported EF value of pond (0.0020) but was same as the value of reservoir in this study.

Compared to the previous results, many studies have also observed lower EFs than the IPCC default value; however, other studies showed higher EFs (Table S1). These inconsistent reports may be reasonable, because a meta-analysis of global data found that riverine N₂O EFs varied across regions (Hu et al., 2016). Meanwhile, our data showed significant negative relationships between EF and N loading (Fig. 4), which is consistent with some previous studies (Beaulieu et al. 2011; Hinshaw and Dahlgren, 2012; Hu et al., 2016). This can be attributed to decreasing microbial activity with increasing N input due to progressive biological saturation (Hu et al., 2016; Mulholland et al. 2008). Although previous studies found that large rivers had low EF (Fu et al., 2018; Hinshaw and Dahlgren, 2012), we proposed that the lower EF may appear in watersheds with a high anthropogenic N input (Fig. 4), highlighting potential constraints in the IPCC methodology for rivers with high N loading. For example, in China, the annual mean riverine N₂O EF in this agriculture watershed with a high N fertilizer rate (~600 kg N ha⁻¹ yr⁻¹) was 0.0013, but in the center of the Sichuan Basin with a relatively low N fertilizer rate (~280 kg N ha⁻¹ yr⁻¹) (Zhou et al. 2012), the value was 0.0027 (Tian et al., 2017).

5. Conclusions

Two years' field measurements in the watershed, an intensively agricultural region located in Eastern China, showed the surface N₂O concentrations and fluxes varied seasonally and spatially. The annual mean surface dissolved N₂O concentrations for rivers, ponds, reservoir, and ditches were 30 nmol L⁻¹, 19 nmol L⁻¹, 16 nmol L⁻¹, and 58 nmol L⁻¹, corresponding to the computed N₂O fluxes of 49.6 μ mol m⁻² d⁻¹, 8.0 μ mol m⁻² d⁻¹, 5.6 μ mol m⁻² d⁻¹, and 45.2 μ mol m⁻² d⁻¹, respectively.

Dissolved N₂O concentration can be best predicted by NO_3^- -N concentrations in rivers (r = 0.71, p < 0.02) and by NH_4^+ in ponds (r = 0.65, p < 0.01). The temporal variation of the riverine N₂O concentration appears to be controlled by precipitation, and heavy precipitation induced the emission hot moments during the farming season. Upstream waters are the hot spots in which riverine N₂O production rates were two times greater than in non-hotspot locations.

The modeled watershed indirect N_2O emission fluxes were comparable to the direct emission fluxes from the fertilized soils. A coarse estimate suggests that indirect N_2O emissions represent approximately 4% of the total N_2O emissions from N-fertilizer at the watershed scale. However, the IPCC default value of 0.0025 may overestimate the agriculture watershed indirect N_2O emission in the study region. We propose that separate EFs for different seasons and different water types may be more appropriate due to the large variability.

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Appendix A. Supplementary data

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