Study of GHGs Flux of water-air Interface and Its Concentration Chang of Lake water, TaiHu

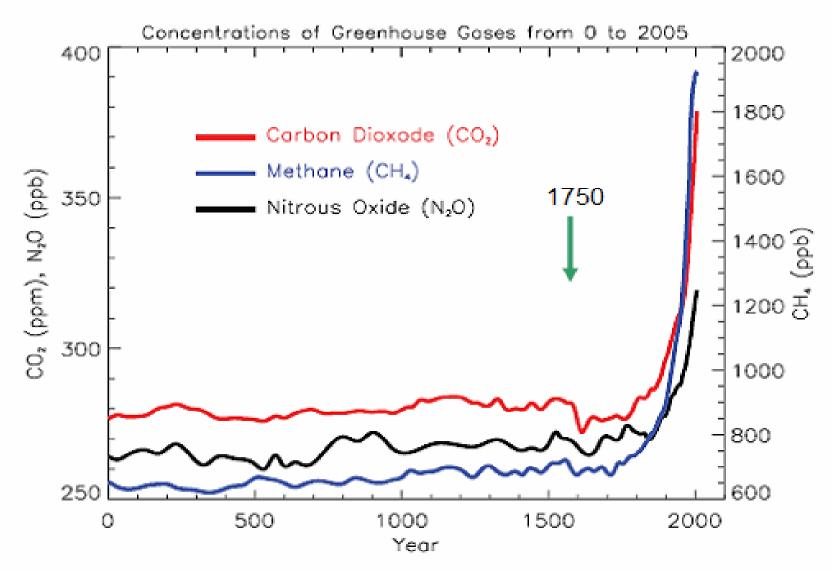
Qitao Xiao

Outline

- 1 Background
- 2 Purpose and Significance
- 3 Content
- 4 Methods
- 5 Innovation
- 6 Progress

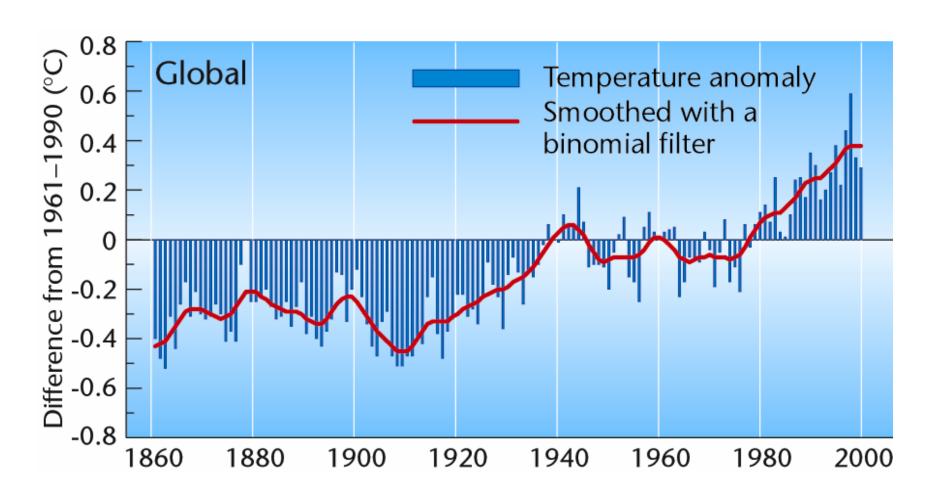
1 Background

Atmospheric concentrations of GHGs over the last 2000 years



Increase since about 1750 are attributed to human activities in the industrial area

Increased concentration of greenhouse gases affect the global temperature



IPCC,2001

Increasing atmospheric concentrations of greenhouse gases CO₂, CH₄, and N₂O have stimulated research on their emission from terrestrial and aquatic environments (Conrad 1996, Segers 1998).

Over the last decade, several field studies conducted worldwide have shown that reservoirs constitute sources of greenhouse gases (GHG) for the atmosphere (Nicolas Soumis et al. 2004).

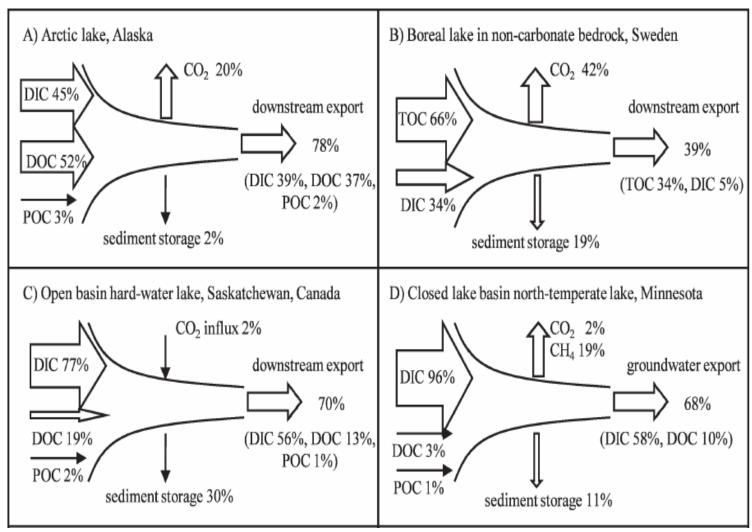
Average fluxes of CO₂ and CH₄from the surfaces of different ecosystems. downward arrows indicates net consumption by ecosystems (sinks) upward arrows indicate net flux to the atmosphere (sources)

Ecosystem	Areal flux (mg · m² · d) ^a				
	CO ₂	CH ₄			
Temperate reservoirs	15001	20↑			
Tropical reservoirs	3000↑	100↑			
Boreal/temperate forests	2100↓	1.0↓			
Tropical forests	710↓	0.2↓			
Northern peatlands	230↓	51↑			
Lakes (worldwide)	700↑	9↑			

Vincent L et al.2000.Bioscience

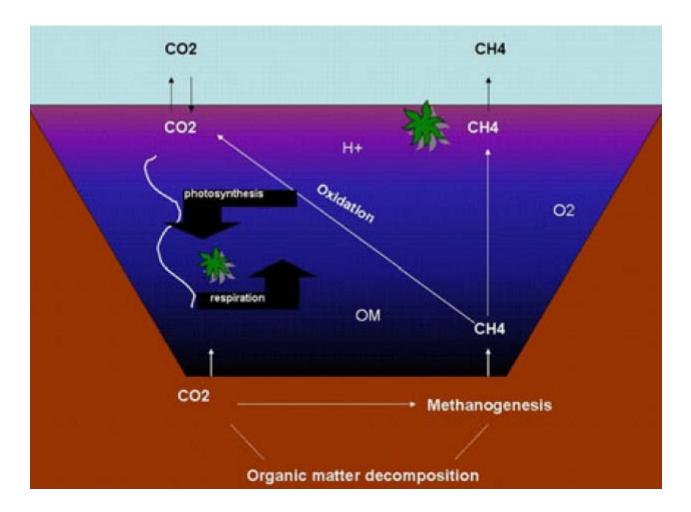
the global C cycle of lake in diverse lakes (by applying the 'active pipe' concep

DIC :dissolved inorganic carbon DOC :dissolved organic carbon POC :particulate organic carbon TOC :total organic carbon



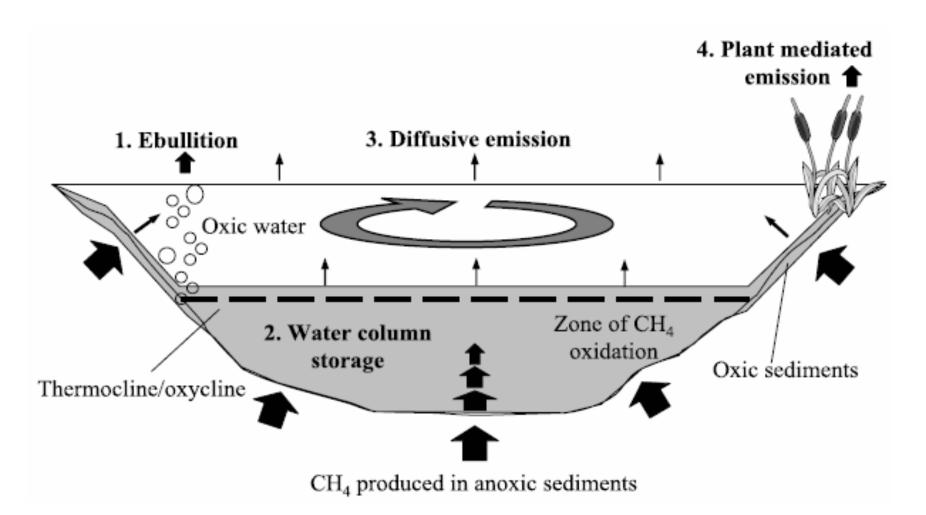
Lars J. Tranvik et al. 2009 .Limnol. Oceanogr

Simplified illustration of CO2 and CH4 dynamics in water bodies (OM organic matter)



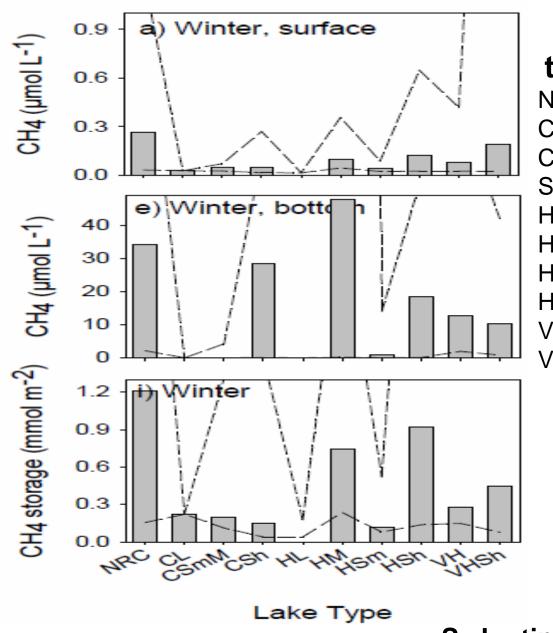
Schrier-Uijl et al. 2011. Biogeochemistry

The emission pathways of GHGs in lake(CH₄, for example)



David Bastviken et al .2004. GLOBAL BIOGEOCHEMICAL CYCLES

The dynamics of GHGs in different lake types (CH4 for example)



the type of the Lake

NRC:Nutrient rich and calcareous

CL: Clear, large

CSmM:Clear, small and middle size

SSh:Clear, shallow

HL: Humic, large

HM: Humic, middle size

HSm:Humic, small

HSH:Humic, shallow

VH:Very humic

VHSH: Very humic, shallow

S. Juutinen et al. 2009. Biogeosciences

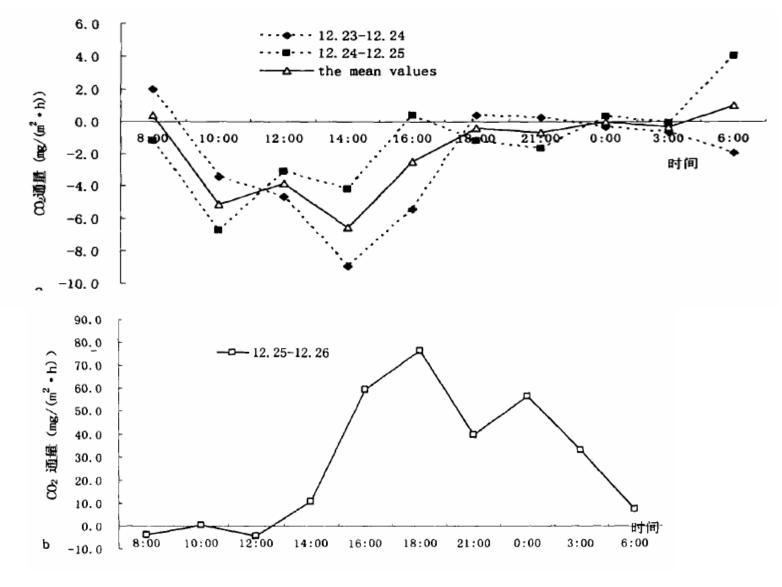
Relative research in China: Lake Taihu

The mean values of diurnal variation in Lake Taihu

CO ₂ flux mg/ (m ² ·h)	Spring	Summer	Autumn	Winter
MLW (sunny)	-0.81	-5.09	-5.29	-1.81
DTH (cloudy)	11.36	-2.45 (sunny)	-	-
Lake Central (cloudy)	5.04	-	-	-

胡维平等. 生态环境. 2004

Diurnal variation of CO2 flux on water-air interface of MLW, Taihu Lake in wintertime



李香华等. 生态学杂志. 2005

2 Purpose and Significance

1).Science assume

Based on relative information, we assumed that Lake Taihu as sources of the GHGs to the atmosphere, and it emissions CO_2 , CH_4 and N_2O from the interface of water-air to atmosphere.

2). The purposes of the study

First, to determine the fluxes of CH_4 , CO_2 and N_2O across the air—water interface of Lake Taihu, a shallow, eutrophic, subtropical lake

Second, to describe the temporal variations of CH_4 , CO_2 and N_2O fluxes and discuss the factors affecting the change

Third, to estimate the C source/sink function of the air—water interface, with discussion on the possible dominant factors in shaping these patterns

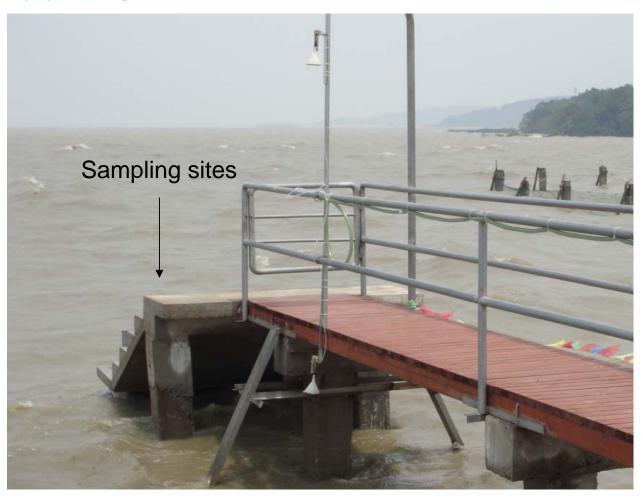
3). The Significance of the study

Accurate, objective evaluation the contribution of the aquatic ecosystems to GHGs and provide data for the estimate of GHGs emission from fresh water bodies.

Provide support for the pollution control of lake.

3 Content

(1)Temporal variation



Sampling at fixed time ,12:00 am every day

(2) Diurnal variation

sampling every three hours

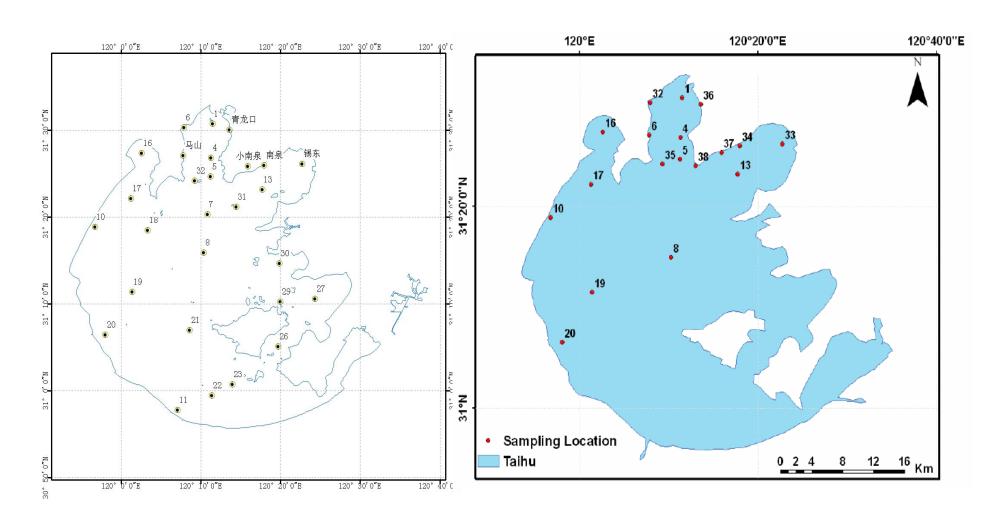
observed for 3 consecutive days

once a month

(3) Vertical profiles

Sampling at 20cm 50cm 100cm 150cm of underwater, respectively

(4) Spatial Variation



2). Study of GHGs flux of Lake Taihu

(1) Temporal variation

(2) Diurnal variation

(3) Spatial Variation

3). Analysis of mechanism

- (1) Relationship between micrometeorology and [GHGx] (Fluxes)
- 1)water temperature (include below 20cm, 50cm, 100cm, 150cm)
- 2 air temperature
- 3 Solar radiation
- 4 wind speed
- **5** wind direction
- **©precipitation**

3). Analysis of mechanism

- (2) Relationship between water quality and [GHGx] (Fluxes)
- ① **pH**
- 2 dissolve oxygen
- ③alkalinity
- **4** Chlorophyll
- **5**Electrical conductance
- **©**Turbidity
- 7The condition of algae bloom
- **®Oxidation-Reduction Potential**

4 Methods

1).Gas Chromatograph method

(1)Sampling

Water samples were collected and poured into glass bottles, without air bubbles, then sealing the stopper with parafilm.

1).Gas Chromatograph method

(2) pre-treatment water samples

Headspace by N₂,100ml Sterilization by CuSO₄ of 2%,5ml-

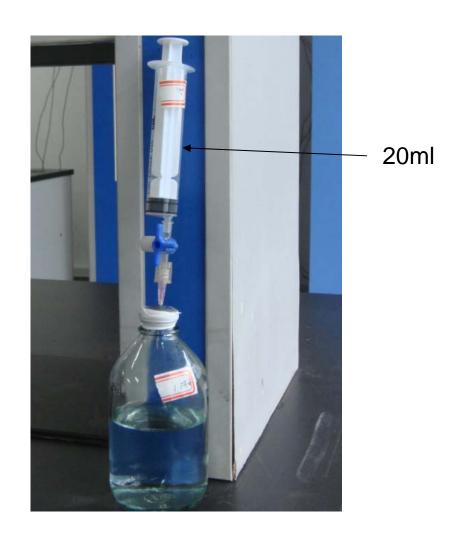
1). Gas Chromatograph method

(3)Gas sampling

the conditions of Gas sampling:

Standing for some time

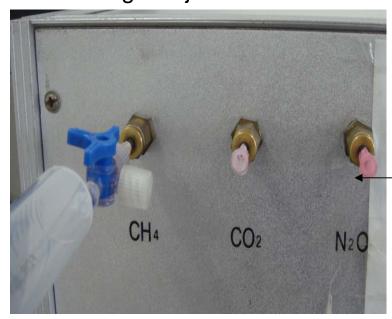
Gas-liquid two-phase equilibrium



1). Gas Chromatograph method

(4) Sampling and analysis

CA-6 gas injection device



the modified Agilent gas chromatograph



Measurement [X]_G (CH4,for example)

 $\frac{\text{CH}_4}{\text{KR}}$ $\frac{\text{CH}_4}{\text{KR}}$ $\frac{\text{KR}}{\text{CH}_4}$ $\frac{\text{KR}}{\text{KR}}$ $\frac{\text$

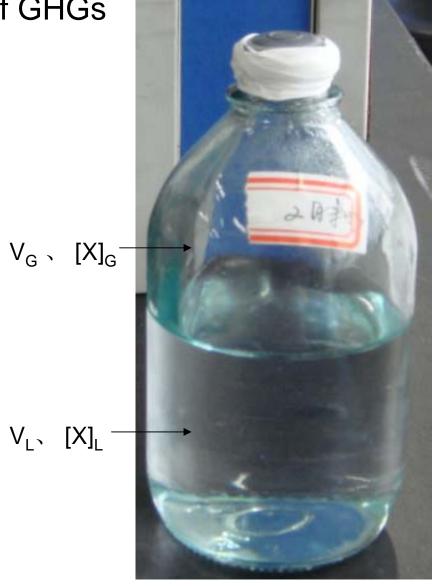
1).Gas Chromatograph method

(4) Calculate the concentration of GHGs

$$[X]_L^0 = [X]_G \times (1/K + \beta)$$

$$K = [X]_G / [X]_L \quad \beta = V_G / V_L$$

[X]_G measured by GC[X]_L calculated by material equilibrium



1) .Gas Chromatograph method

(5) The study of flux

the flux was calculated using the following equations:

$$F = k (Cw - Ceq) \dots (1)$$

where k is a piston velocity that was calculated by

$$k/k_{600} = (Sc/Sc_{600})^{(-n)}$$
..... (2)
 $U_{10} < 3.7 \text{m/s } n = 2/3$ $U_{10} > 3.7 \text{m/s } n = 0.5$

$$U_{10}/U = \ln(10/Z_0)/\ln(3.5/Z_0)$$
 (3)

where Z_0 is roughness $Z_0 = 0.00035$ m where U refers to the wind speed at the 3.5-m height

keoo is the corresponding value for Sc600

$$k_{600} = 2.07 + 0.215U_{10}$$
 (4)

2).Picarro CO2、CH4 analyzer

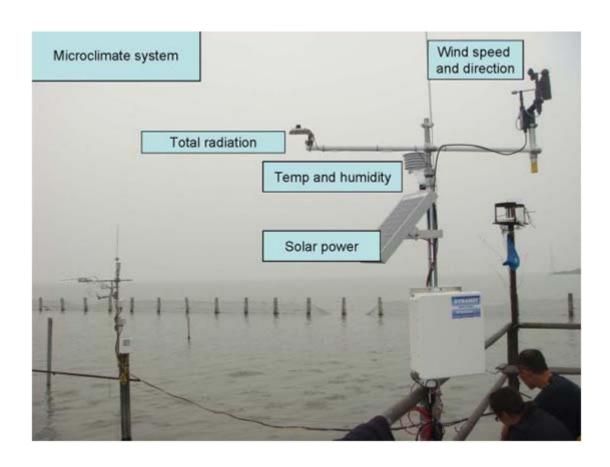


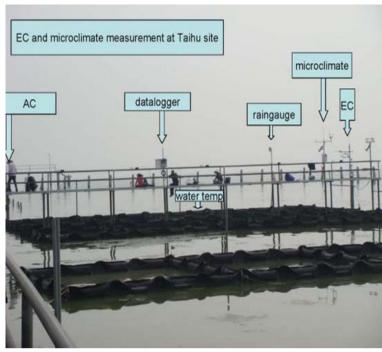




Sampling port

3). Access to the data of meteorological





4). Access to the data of water quality

太湖站水华预警监测现场测定结果

			采样时间: 2010年 5 月 27 日				7 日		
水华	采样时间	水温℃	电导率Sp/Cond (ms/cm)	рН	ORP (mV)	浊度 (MTV)	Chlorphy (ug/L)	BGA (cells/mL)	DO (mg/L)
有蓝藻颗粒	6:51	22.76	0. 536	7. 92	442.7	19.5	5.7	2304	8.24
有蓝藻颗粒	7:17	22.9	0.524	7. 47	472.7	20.2	13.1	3209	6
有蓝藻颗粒	7:24	22.86	0.627	7.66	386.3	12.5	2.4	1925	7. 29
有蓝藻颗粒	7:48	23.31	0.67	7.6	516.8	45.0	21.2	4675	6.67
有蓝藻颗粒	8:20	22.52	0. 493	8.06	437.8	27.1	3.9	1963	8. 42
有蓝藻颗粒	8:33	22.49	0. 488	8.02	444.7	42.9	4.9	2526	8.34
有蓝藻颗粒	8:56	22. 4	0. 507	8.05	444.1	53.6	4.6	2467	8. 45
有蓝藻颗粒	9:33	23.34	0. 525	8.07	418.8	11.1	1.1	1582	8. 46

5 Innovation

The innovation of the study

1.GC combined with Picarro

2. Synchronous study the CO2 CH4 and N₂O

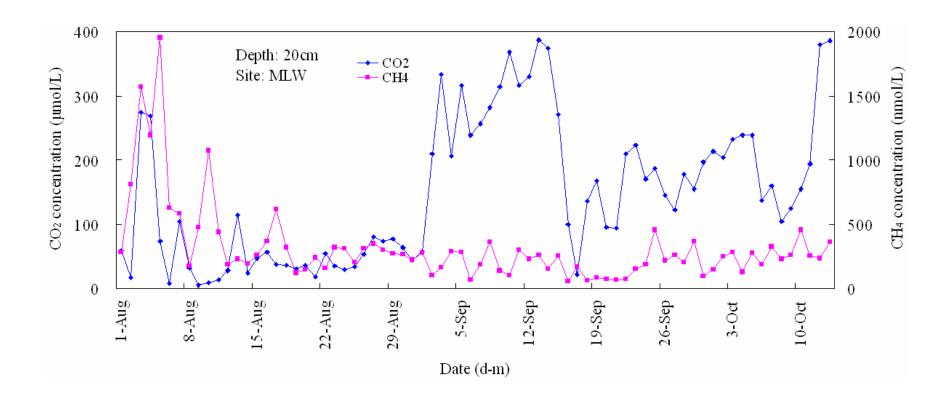
The innovation of the study

3.Research related to freshwater lakes is mostly concentrated in the small and medium lakes which productive is not high, deep lakes, ecological and hydrological characteristics of simple types, and the study of medium-sized lake which affected by human activities is more complex has not been reported in the literature.

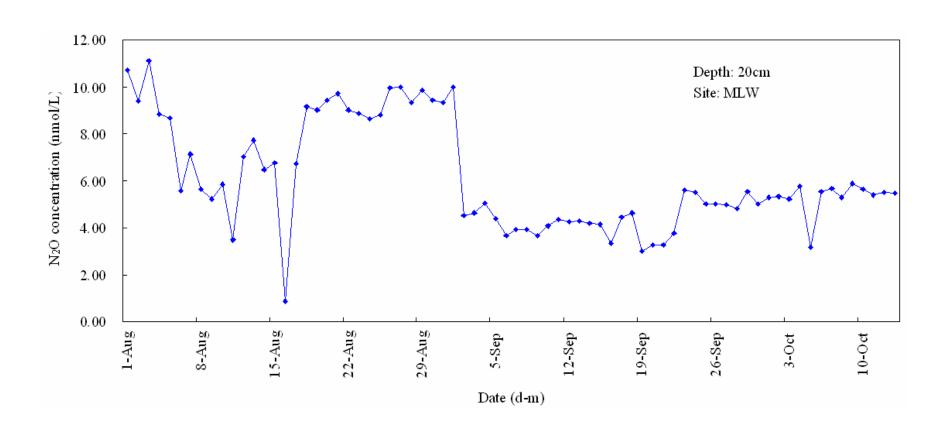
The study of Lake Taihu has a strong representation

6 Progress

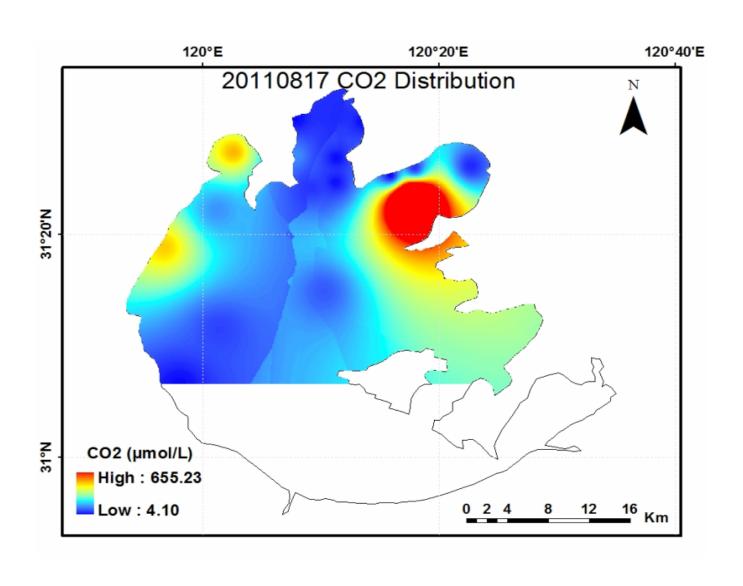
1). Temporal variation



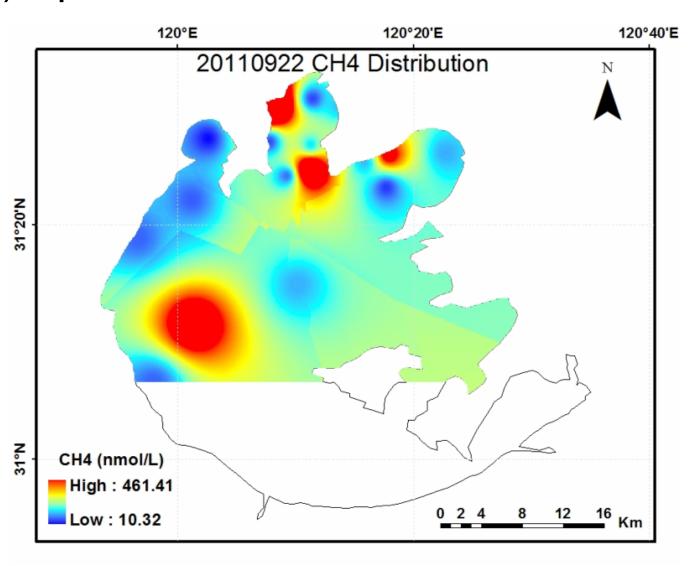
1).Temporal variation



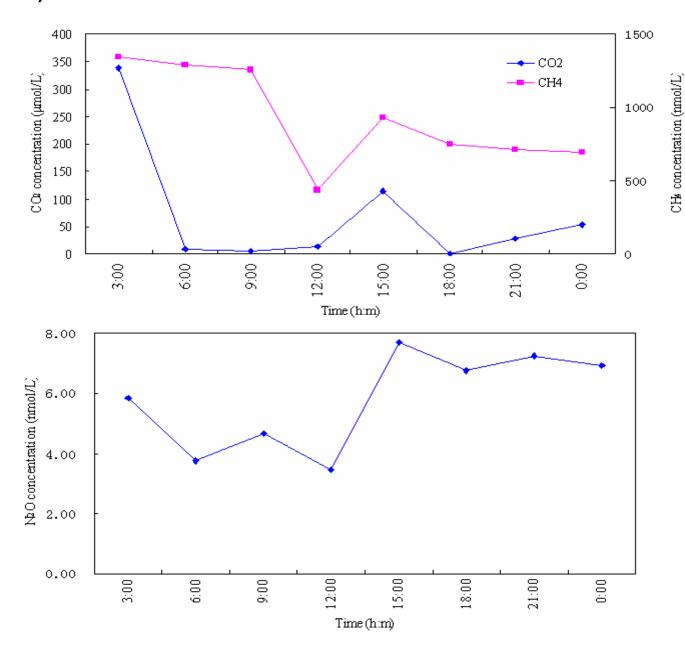
2). Spatial variation



2). Spatial variation

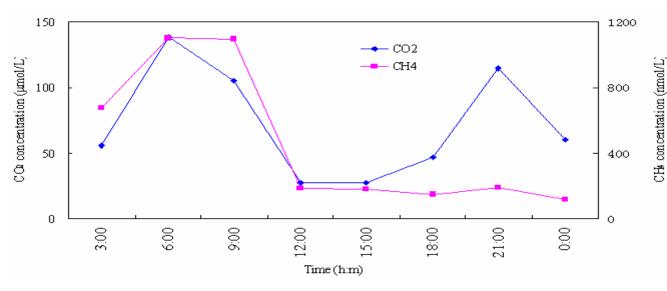


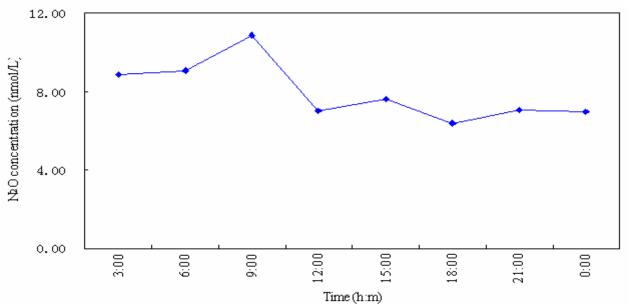
3). Diurnal variation



11 August, 2011 Site:MLW

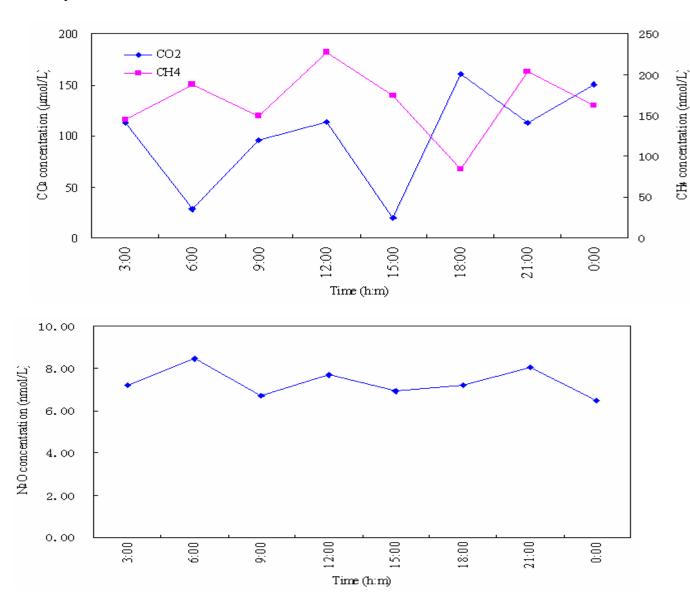
3). Diurnal variation





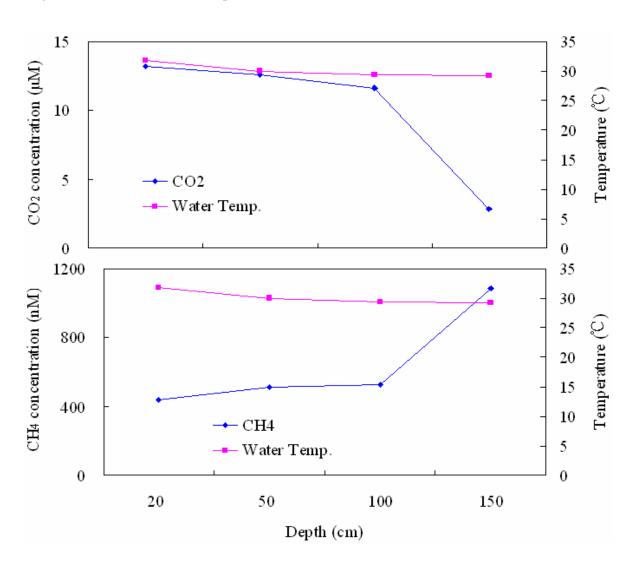
12 August, 2011 Site:MLW

3). Diurnal variation



13 August, 2011 Site:MLW

4). Vertical profiles and Micromet



24:00 11 August, 2011 Site:MLW

5).Fluxes of GHGs

