

# **Improving the perform of a low-cost CO<sub>2</sub> monitoring system for urban research**

# OUTLINE

## Brief introduction

## Specific experiments are as follows

1. Calibration in the lab.
2. Calibration during the urban CO<sub>2</sub> observation.
3. Tank gas observation(416ppm CO<sub>2</sub>) .
4. Observation in buffer.
5. Atmosphere observations compare compared with Picarro.

## conclusion

## Brief introduction

### Objective

We want to observe the temporal and spatial distribution of CO<sub>2</sub> concentration in Nanjing urban area.

### Challenge

The most important factor influencing the instrument measurement is temperature. LI-840A has a self heating system, but only for the optical cavity. Optical cavity temperature will change if the rapid change of ambient temperature and rapid air flow happen, and then make the measurement inaccurate.

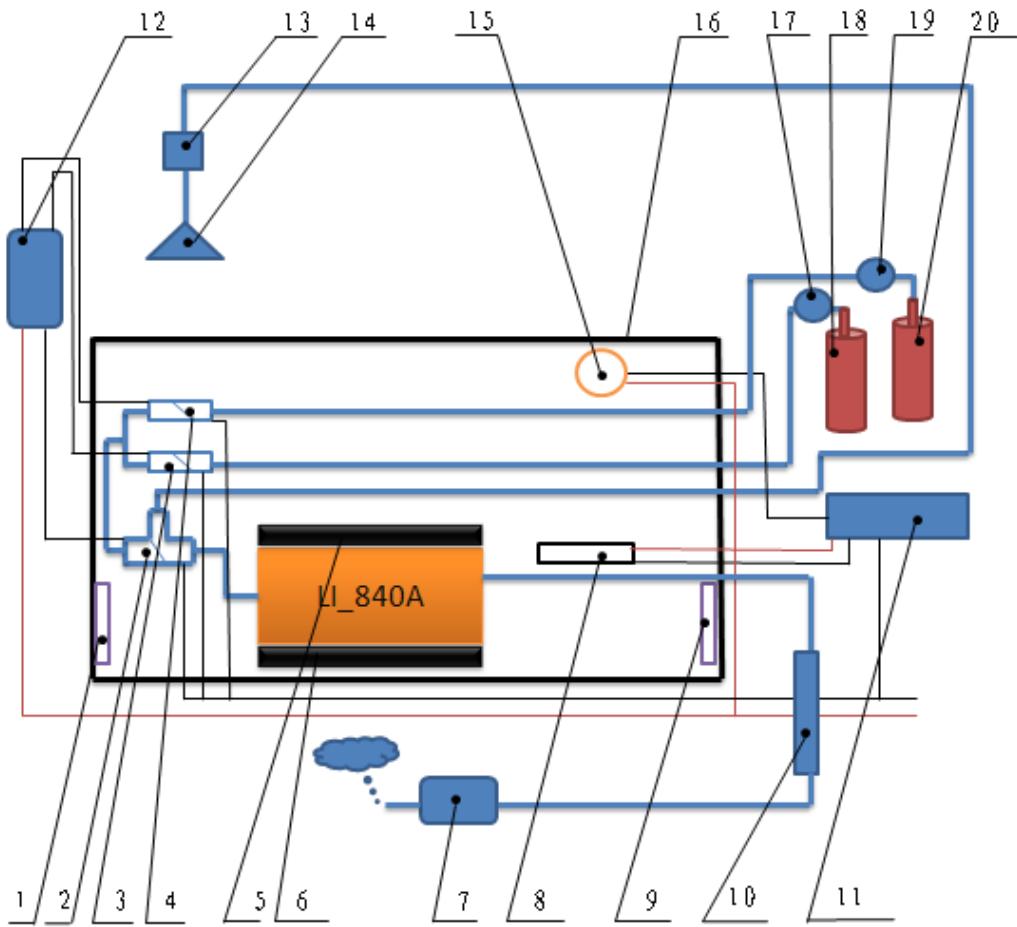
Instrumental measurements have great drift, We need to know the calibration frequency.

### The box

Function : Heating insulation; automatic calibration.

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1	fan
2	electromagnetic valves
3	electromagnetic valves
4	electromagnetic valves
5	Metal plate
6	Metal plate
7	Pump
8	Temperature sensor
9	Fan
10	Flow meter (500cc/min)
11	Temperature control circuit
12	Calibration controller
13	Filter
14	Rain shield
15	Heater
16	Case
17	Reducing valve
18	Standard gas
19	Reducing valve
20	Standard gas

## The cost of the box

Part	Price	Amount
SOLENOID VALUES	\$ 35x3	\$ 105
PUMP	\$ 300X1	\$ 300
FLOWMETER	\$ 40X1	\$ 40
CROSS UNION	\$ 13.6X4	\$ 54.4
DEKABON TUBING	\$ 0.66x10	\$ 6.6
<b>BOX</b>	<b>500 ¥ x1</b>	<b>\$ 82.2</b>
<b>Value and temperature control unit</b>	<b>100 ¥ x1</b>	<b>\$ 16.4</b>
<b>other</b>	<b>200 ¥</b>	<b>\$ 32.9</b>
Bought in America		\$ 506
<b>Bought in China</b>		<b>\$ 131.5</b>
<b>TOTAL</b>		<b>\$ 637.5</b>

# 1. Calibration in the lab

We calibrated all the instruments before urban CO<sub>2</sub> observation.

Table.1 Calibration on June 4<sup>th</sup>

	Zero CO <sub>2</sub>	490ppmCO <sub>2</sub>	Zero H <sub>2</sub> O	22°C H <sub>2</sub> O	1°C H <sub>2</sub> O
LI-840A_No. 1	-1.4ppm	480.6ppm	-0.00017ppt	26.98ppt	6.45ppt
LI-840A_No. 2	-0.4ppm	478.6 ppm	-0.00085ppt	28.22ppt	7.92ppt
LI-840A_No. 3	-0.06ppm	476.0ppm	0ppt	28.17ppt	6.33ppt
LI-840A_No. 4	-7.6ppm	488.2ppm	-0.001ppt	27.56ppt	6.34ppt
LI-840A_No. 5	-2.7ppm	498.1ppm	-0.00033ppt	29.19ppt	6.00ppt
<b>standard</b>	<b>0ppm</b>	<b>486.1ppm</b>	<b>0ppt</b>	<b>26.54ppt</b>	<b>6.60ppt</b>

**We did not test the 490ppm CO<sub>2</sub> standard gas at that time, the true concentration of the gas is 486.1ppm CO<sub>2</sub>.**

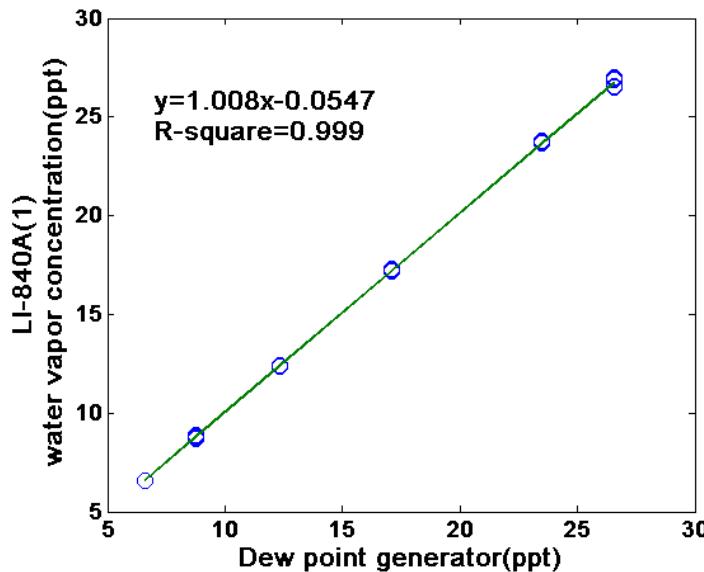


Fig.1

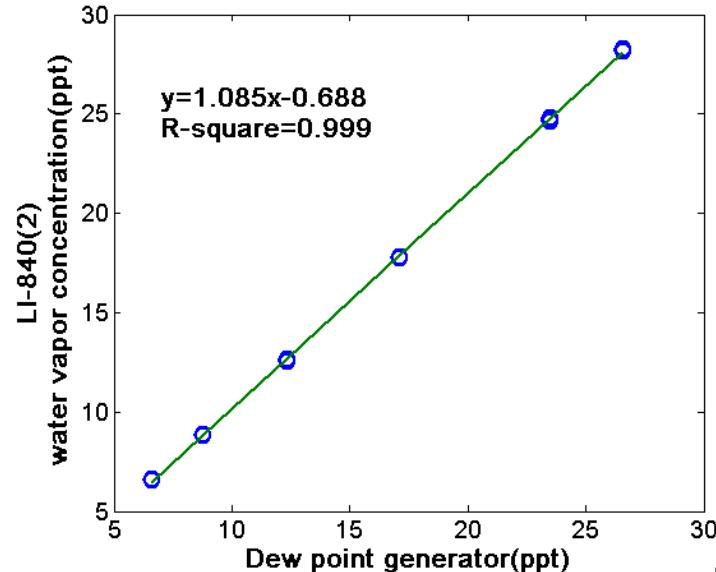


Fig.2

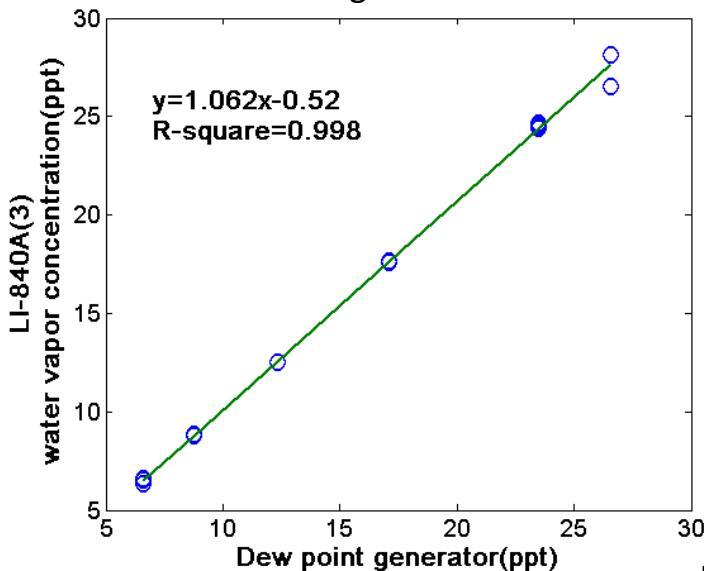


Fig.3

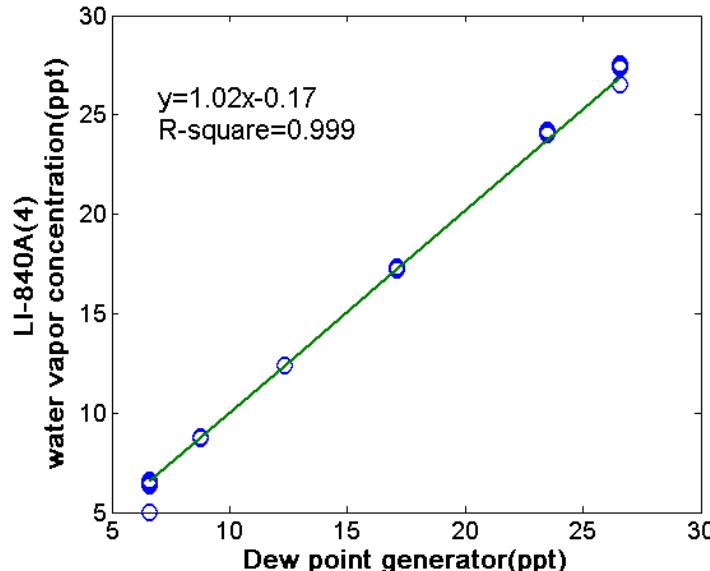


Fig. 4

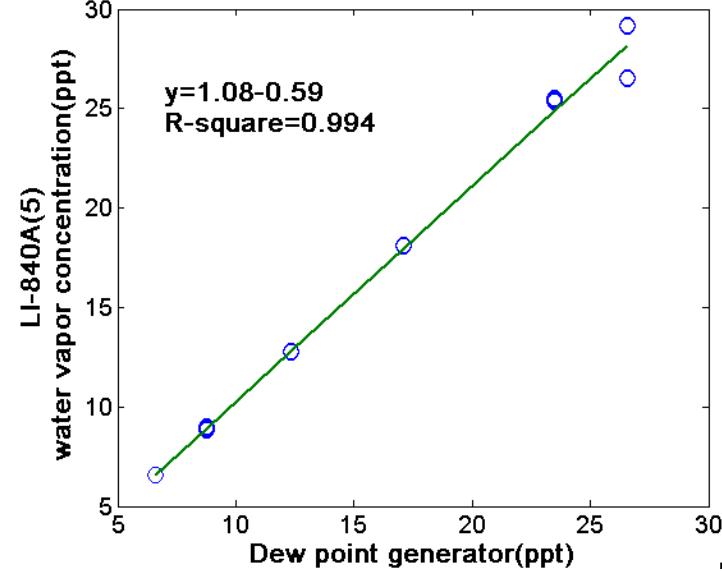


Fig. 5

Table.2 Before calibration water vapor concentration

	1°C	5°C	10°C	15°C	20°C	22°C )
LI-840A_No. 1	6.45ppt	8.76ppt	12.34ppt	17.23ppt	23.79ppt	26.98ppt
LI-840A_No. 2	7.92ppt	8.82ppt	12.60ppt	17.74ppt	24.77ppt	28.22ppt
LI-840A_No. 3	6.33ppt	8.80ppt	12.53ppt	17.65ppt	24.69ppt	28.17ppt
LI-840A_No. 4	6.34ppt	8.78ppt	12.38ppt	17.36ppt	24.19ppt	27.56ppt
LI-840A_No. 5	6.00ppt	8.88ppt	12.76ppt	18.10ppt	25.51ppt	29.19ppt
Standard vapor concentration	6.60ppt	8.76ppt	12.33ppt	17.12ppt	23.47ppt	26.54ppt

## 2. Calibration during urban CO<sub>2</sub> observation

Urban CO<sub>2</sub> observation lasted from June 6<sup>th</sup> to June 23<sup>th</sup>.

Only four instruments was used during the 5 times observation with calibration.

Table.3 Calibration during the urban CO<sub>2</sub> observation

	Zero CO <sub>2</sub> (ppm)			490ppmCO <sub>2</sub> (ppm)			Zero H <sub>2</sub> O (ppt)		
Five times calibration	Average	min	max	Average	min	max	Average	min	max
LI-840A_No. 1	-0.87	-0.25	-1.5	488.43	486.2	490.5	0.14	-0.01	-0.68
LI-840A_No. 3	-0.64	-0.1	-1.66	488.75	487.2	490.2	0.10	0.03	0.31
LI-840A_No. 4	-1.08	0.3	-2.52	488.20	486.5	490.2	0.20	-0.03	-0.75
LI-840A_No. 5	-0.07	1.01	-4.15	490.364	485.3	496.6	0.06	-0.03	0.9
<b>standard</b>	<b>0ppm</b>			<b>486.1ppm</b>			<b>0ppt</b>		

Table.4 The last calibration during urban CO<sub>2</sub> observation was on June 23<sup>rd</sup>

Standard gas: N <sub>2</sub>	CO <sub>2</sub> before calibration	CO <sub>2</sub> after calibration	H <sub>2</sub> O before calibration	H <sub>2</sub> O after calibration
LI-840A_No. 1	-0.5ppm	0	-0.02ppt	0
LI-840A_No. 2	-4.5ppm	0	0.02ppt	0
LI-840A_No. 3	-0.1pm	0	0.05ppt	0
LI-840A_No. 4	-1.5ppm	0	-0.04ppt	0
LI-840A_No. 5	-2.1ppm	0	-0.03ppt	0
<b>Standard gas: 490ppm CO<sub>2</sub></b>	<b>CO<sub>2</sub> before calibration</b>	<b>CO<sub>2</sub> after calibration</b>		
LI-840A_No. 1	486.2ppm	490ppm		
LI-840A_No. 2	500.0ppm	490ppm		
LI-840A_No. 3	487.5ppm	490ppm		
LI-840A_No. 4	487.5PPm	490ppm		
LI-840A_No. 5	485.3ppm	490PPm		

## Experience

LI-840A was in the worst conditions during urban CO<sub>2</sub> observation. Violent vibration in the car, switching power frequently and the change of voltage make the instrument performance not satisfactory.

In the June 13<sup>th</sup> 490ppm CO<sub>2</sub> calibration, we found large gap between instruments.

But the rest of the time, the difference of CO<sub>2</sub> was less than  $\pm 2\text{ppm}$ .

The LI-840A\_NO.5 performance is the worst, we need to pay attention to it.



### 3. Tank gas observation(416ppm CO<sub>2</sub>)

The observation started from June 23<sup>rd</sup> to the end of June 26<sup>th</sup>, lasted 58 hours. We are concerned about the error between instruments. 416ppm is nominal concentration, we do not know the true concentration of the gas.

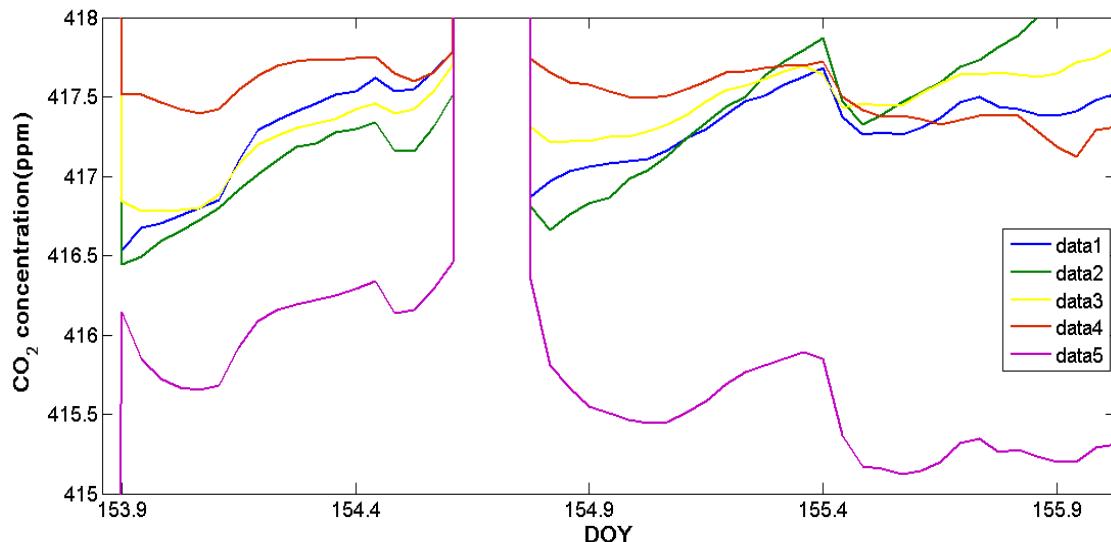


Fig.6 Tank gas observation, hourly average of CO<sub>2</sub> concentration.

Table.5 The mean value of tank gas observation.

Instruments	Mean CO <sub>2</sub>
LI-840A_No. 1	417.2ppm
LI-840A_No. 2	417.0ppm
LI-840A_No. 3	417.1ppm
LI-840A_No. 4	417.6ppm
LI-840A_No. 5	416.0ppm

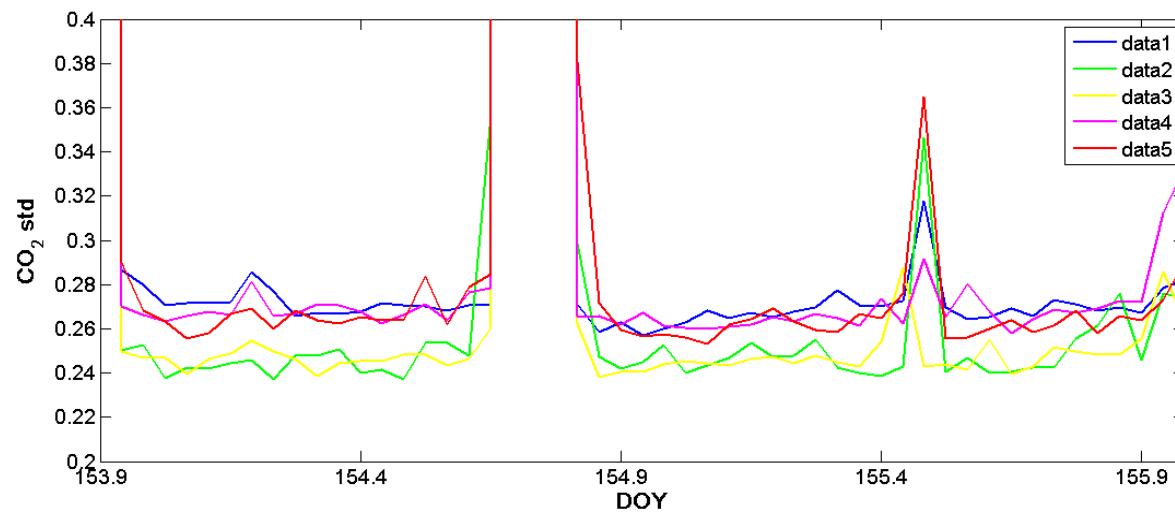


Fig.7 Hourly CO<sub>2</sub> concentration standard deviation .

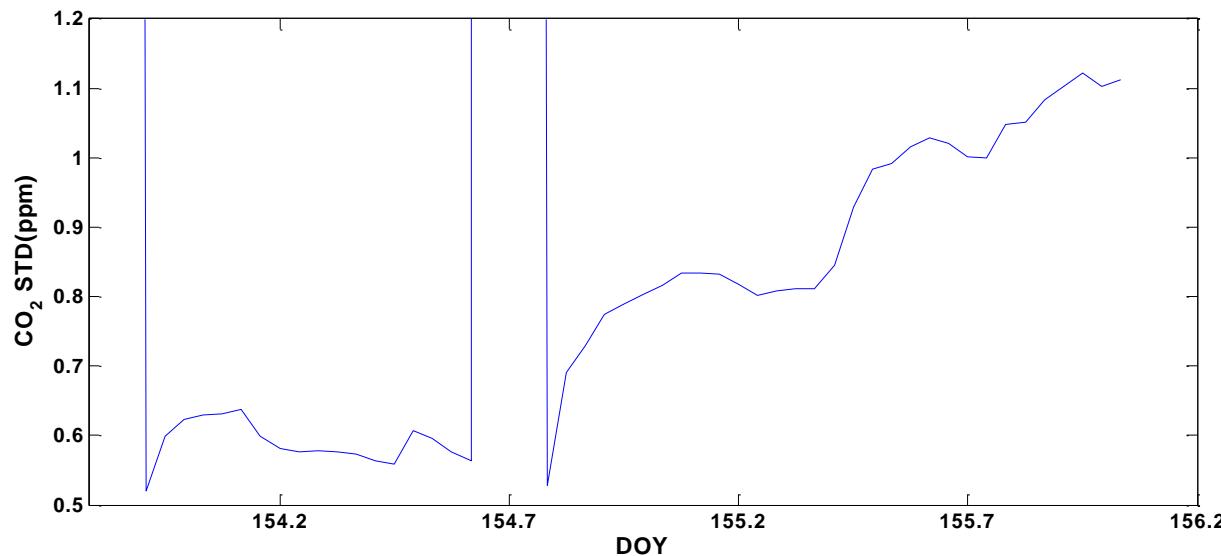


Fig.8 CO<sub>2</sub> concentration standard deviation among instruments.

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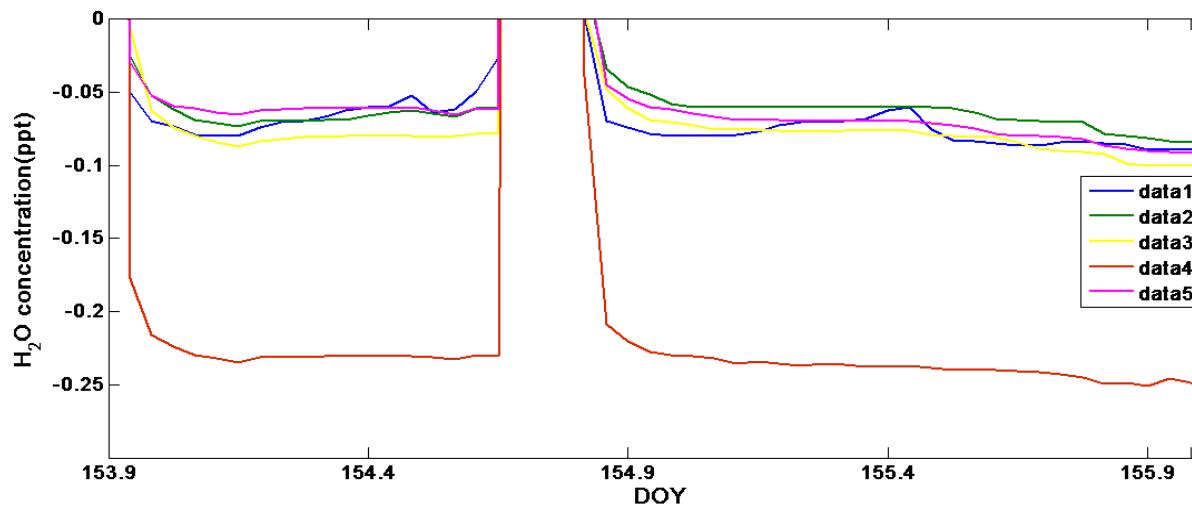


Fig.9 Tank gas observation, hourly H<sub>2</sub>O concentration.

Table.6 The mean value of H<sub>2</sub>O in tank gas observation.

Instruments	Mean H <sub>2</sub> O
LI-840A_No. 1	-0.061ppt
LI-840A_No. 2	-0.067ppt
LI-840A_No. 3	-0.08ppt
LI-840A_No. 4	-0.23ppt
LI-840A_No. 5	-0.06ppt

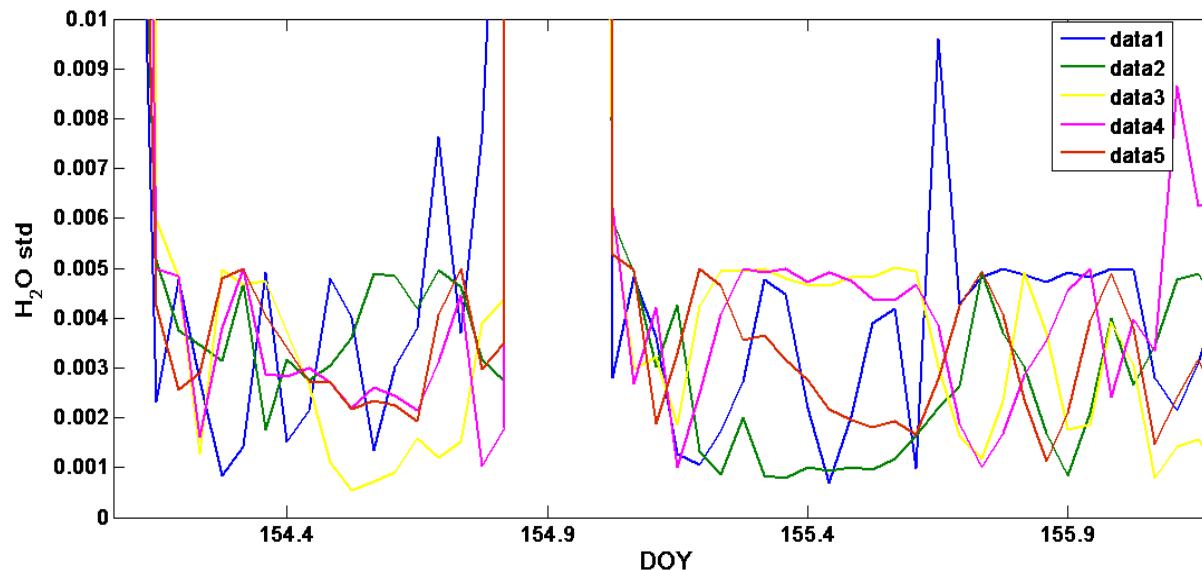


Fig.10 H<sub>2</sub>O concentration standard deviation of hourly average.

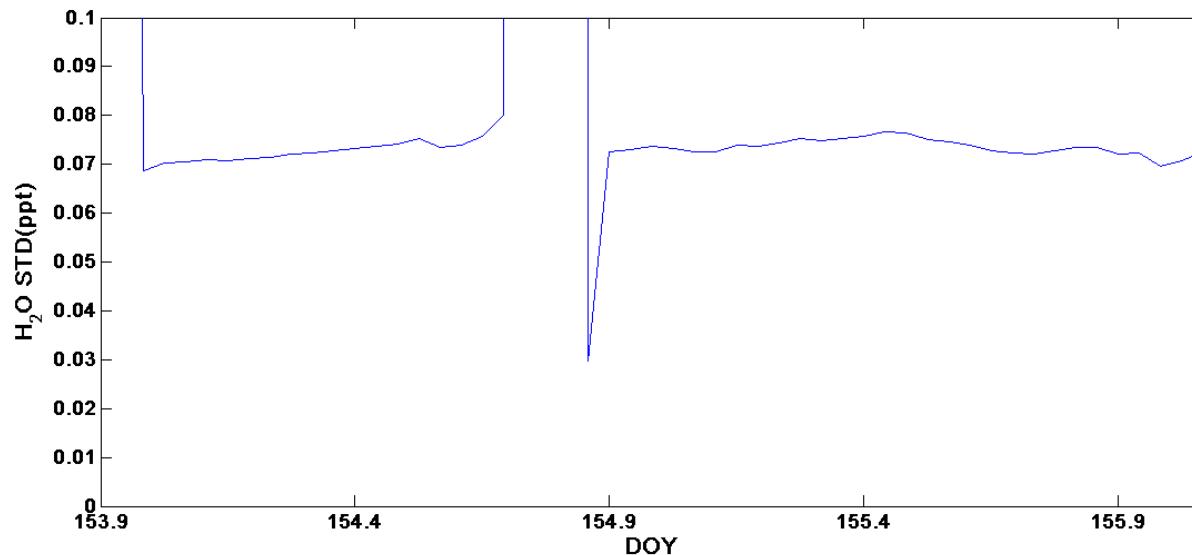


Fig 11.  $H_2O$  concentration standard deviation among instruments..

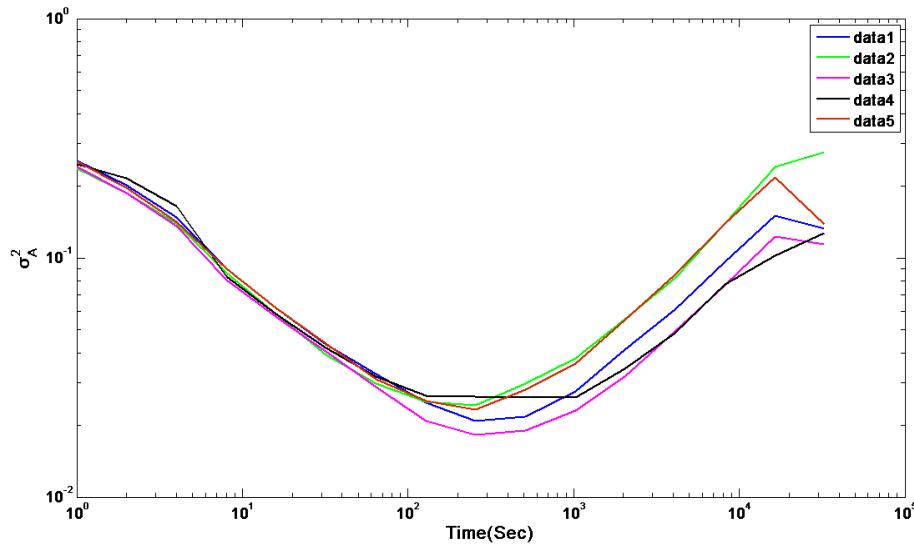


Fig12. Alan variance of  $CO_2$

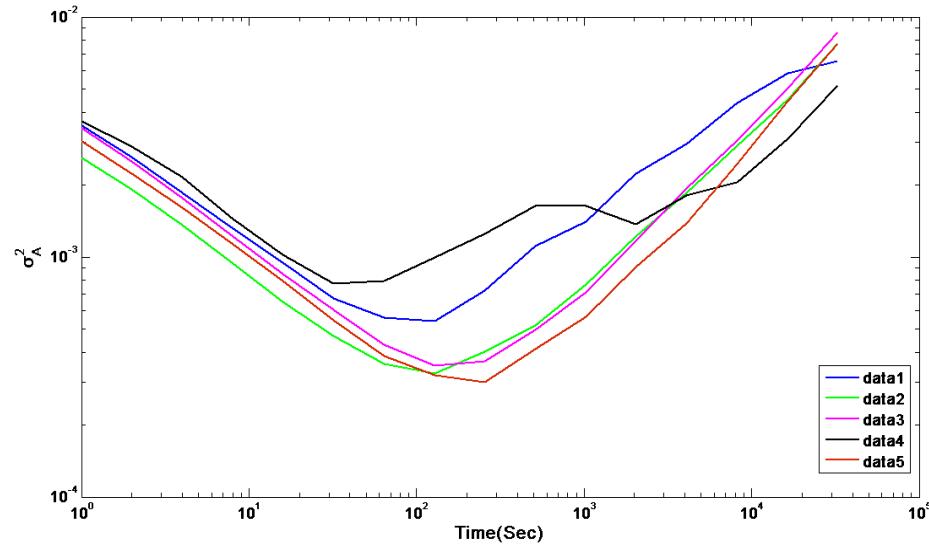


Fig13. Alan variance of  $H_2O$

## **Experience**

This experiment was carried out in the most ideal environment.

Water vapor measuring is stable while the drift of CO<sub>2</sub> measurement is obvious.

Drift of CO<sub>2</sub> measurement is about 1ppm per day and accumulates over time. The drift may be unacceptable if the instrument works more than three days without calibration.

CO<sub>2</sub> measurement of instrument No. 5 and water vapor measurement of instrument No. 4 seems different from the others.

## 4. Observation in buffer

Buffer observation started on June 27<sup>th</sup>, four days after calibration, followed by the 416ppm CO<sub>2</sub> stable gas observation.

Observation lasted for 6days. All data correction use July 3<sup>rd</sup> calibration data, and water vapor correction were also conducted.

When we conducted calibration in July 3<sup>rd</sup>, LI-840\_NO.3 has something wrong so that we must restore factory default settings. As it's parameters of the calibration and observation were different, the NO.3 CO<sub>2</sub> data is not displayed.

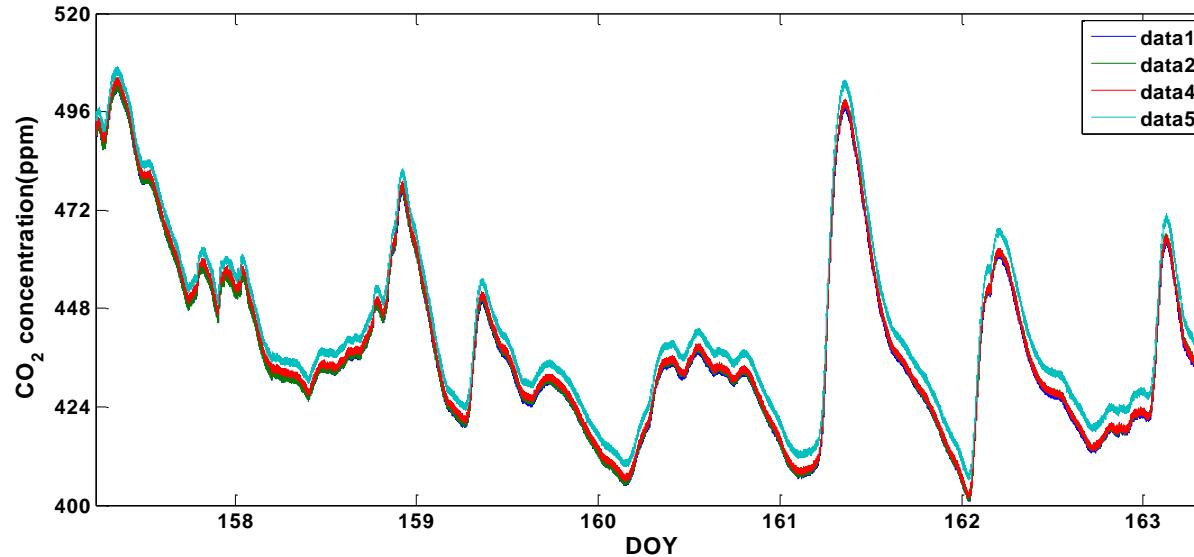


Fig.13 Observation in buffer, dry  $\text{CO}_2$  concentration time series.

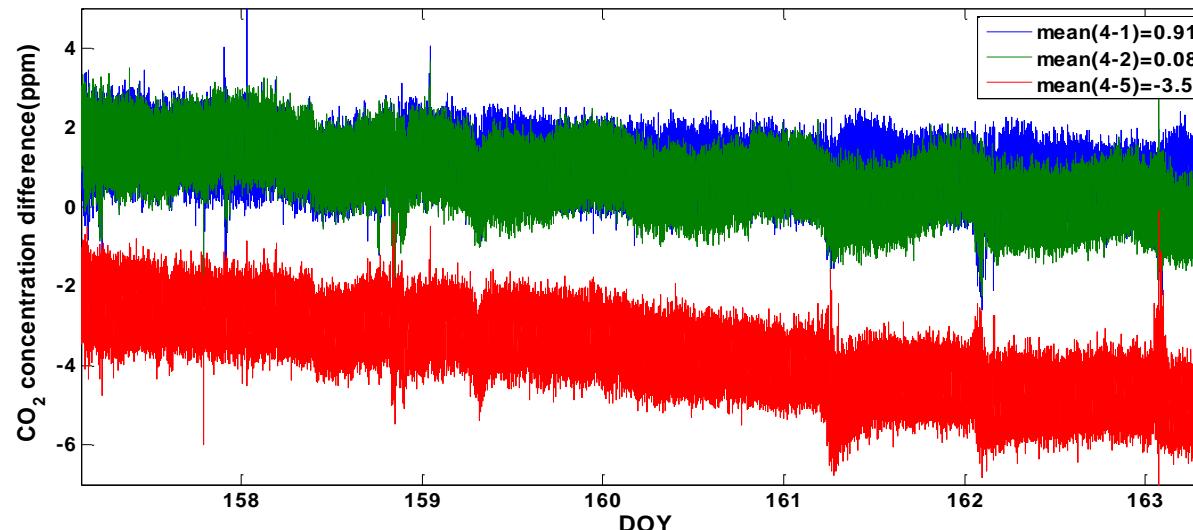


Fig.14 Observation in buffer, dry  $\text{CO}_2$  concentration difference.

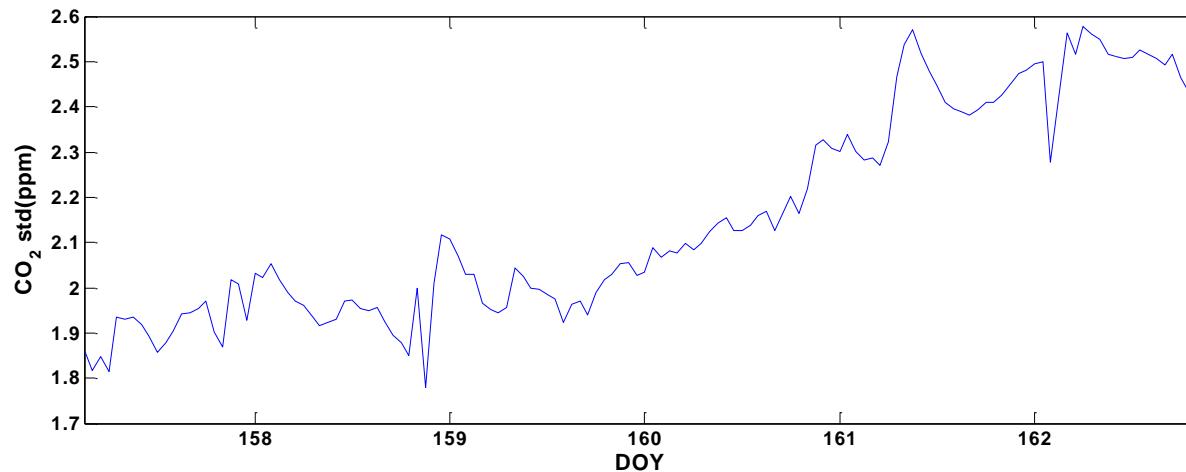


Fig.14 Observation in buffer, dry CO<sub>2</sub> concentration standard deviation of hourly average.

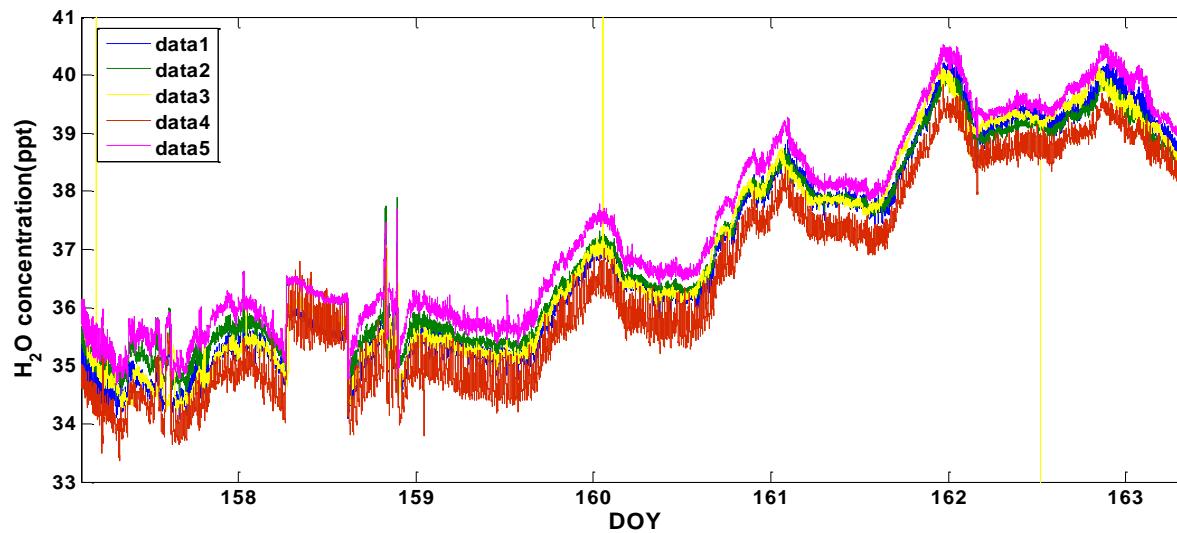


Fig.15 Observation in buffer,  $\text{H}_2\text{O}$  concentration time series.

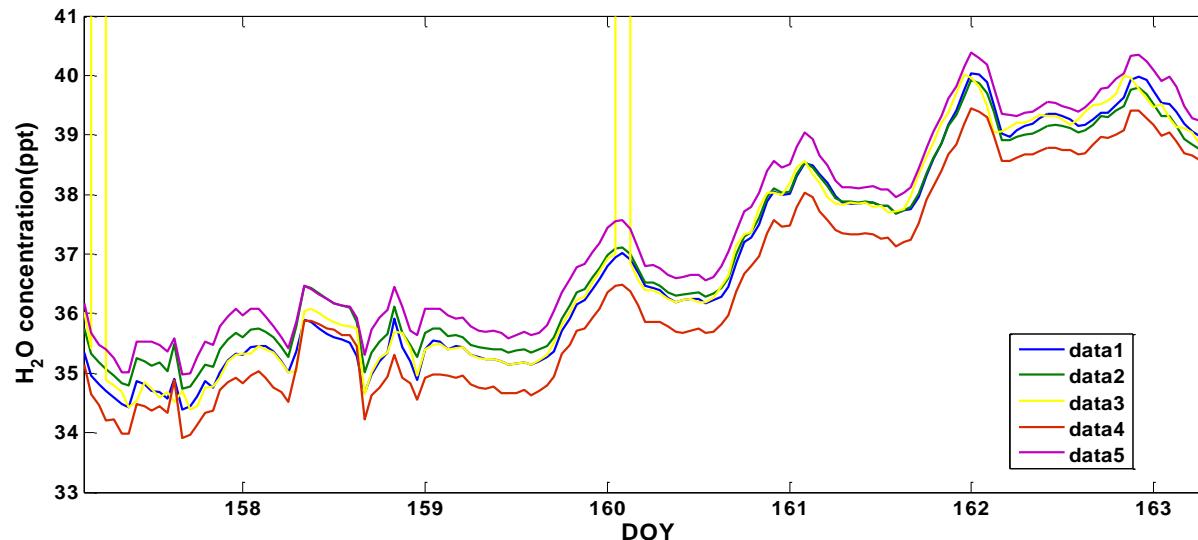


Fig.16 Observation in buffer, dry  $\text{CO}_2$  concentration hourly average.

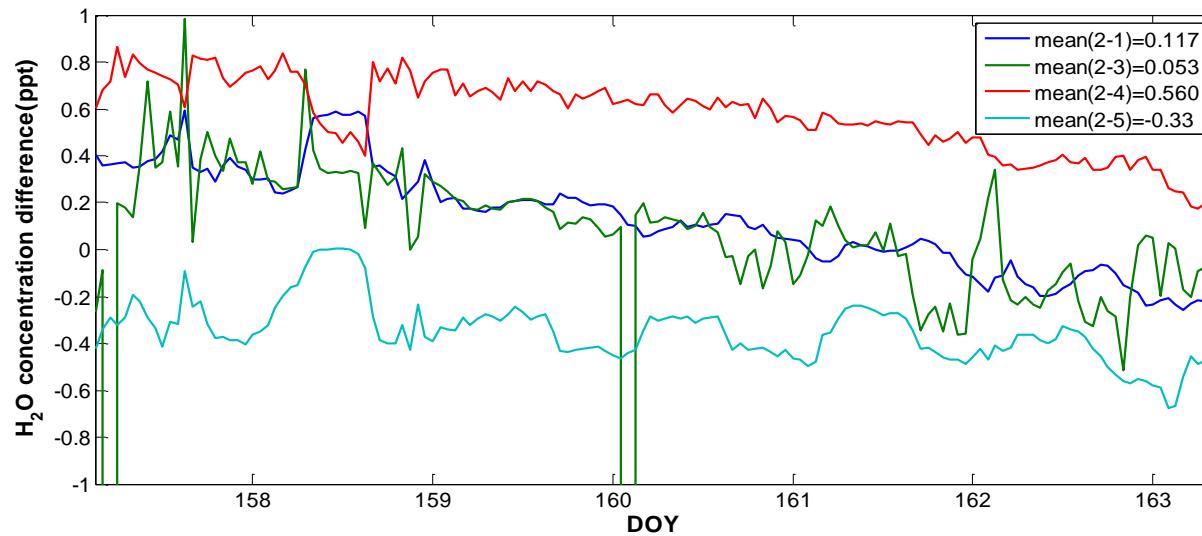


Fig. 15 Observation in buffer,  $H_2O$  concentration difference.

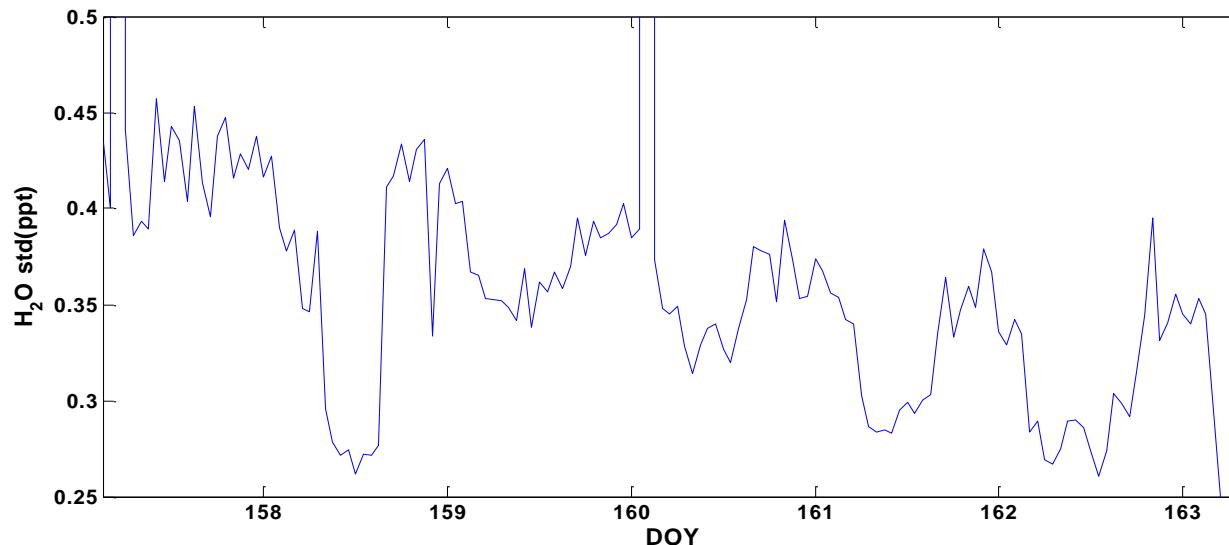


Fig.16  $H_2O$  concentration standard deviation among instruments.

## **Experience**

Combined with the last experiment, the measurement of water vapor concentration is stable though the water vapor concentration variation range is large.

There are obvious noise for LI-840A .

Factory setting is an important data in the maintenance process.



## 5. Atmosphere observations compared with picarro.

The experiment lasted from July 5<sup>th</sup> to July 25<sup>th</sup>.

We carried out calibration on July 2<sup>nd</sup> and July 21<sup>st</sup>.

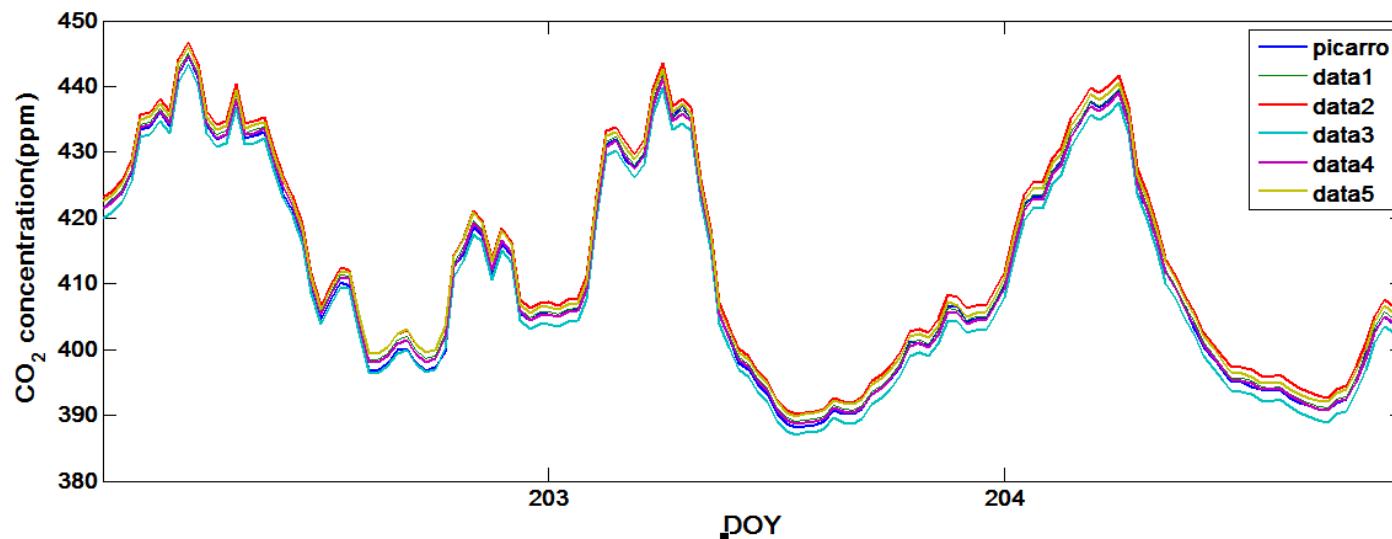


Fig.17 Atmosphere observations, hourly dry CO<sub>2</sub> concentration.

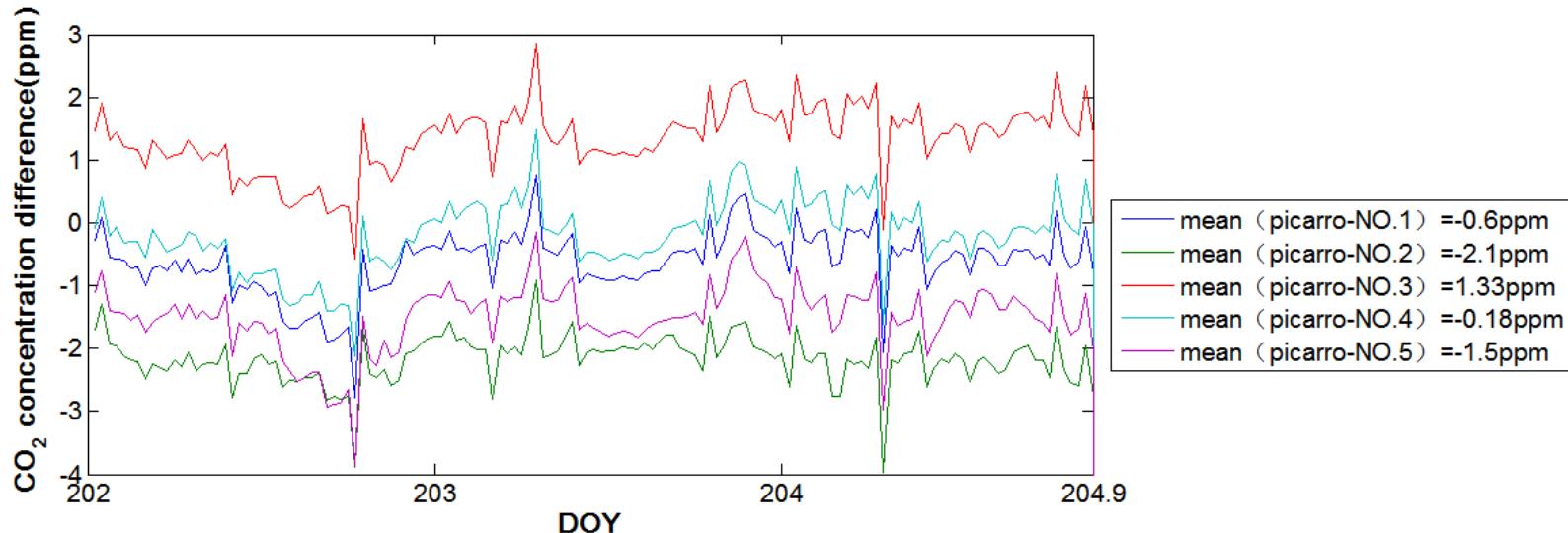


Fig.18 Atmosphere observations, hourly dry CO<sub>2</sub> concentration difference.

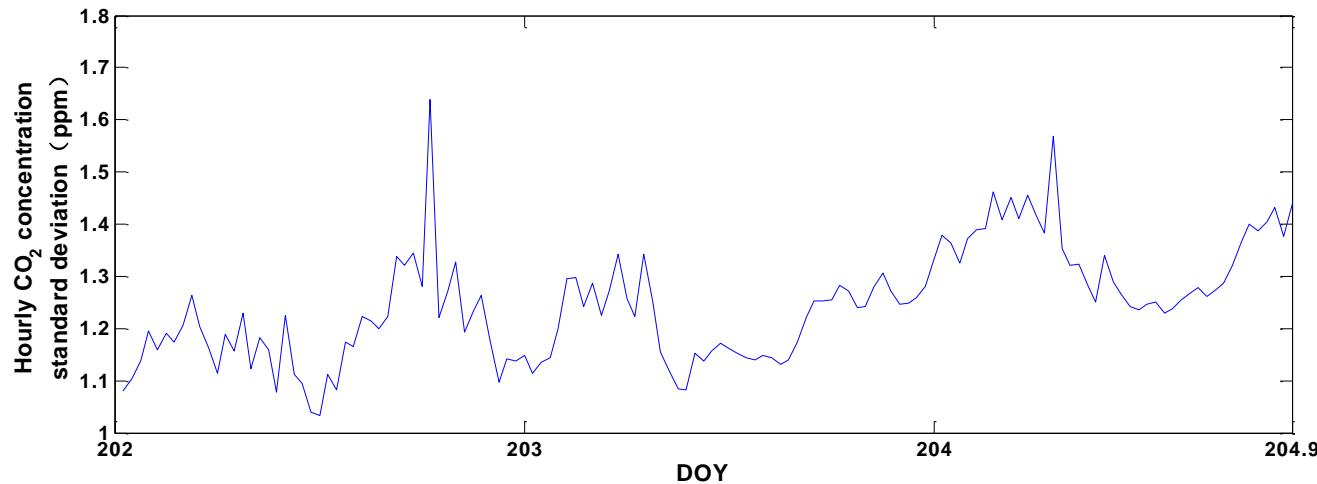


Fig.19 Atmosphere observations ,hourly CO<sub>2</sub> concentration standard deviation.

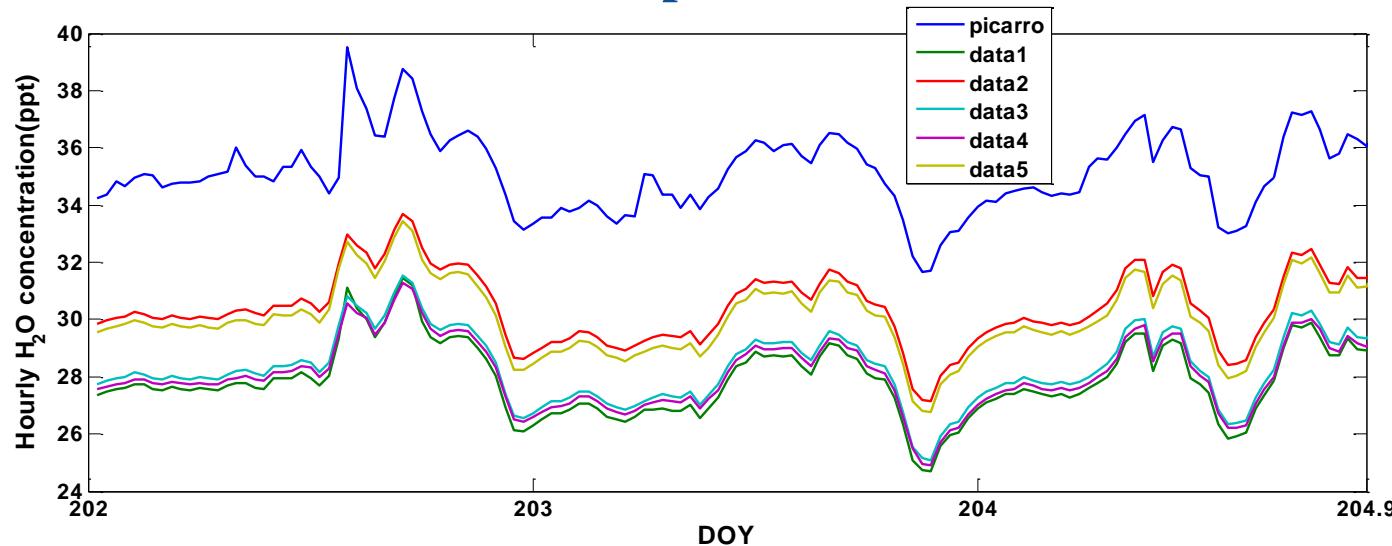


Fig.20 Atmosphere observations, hourly  $\text{H}_2\text{O}$  concentration.

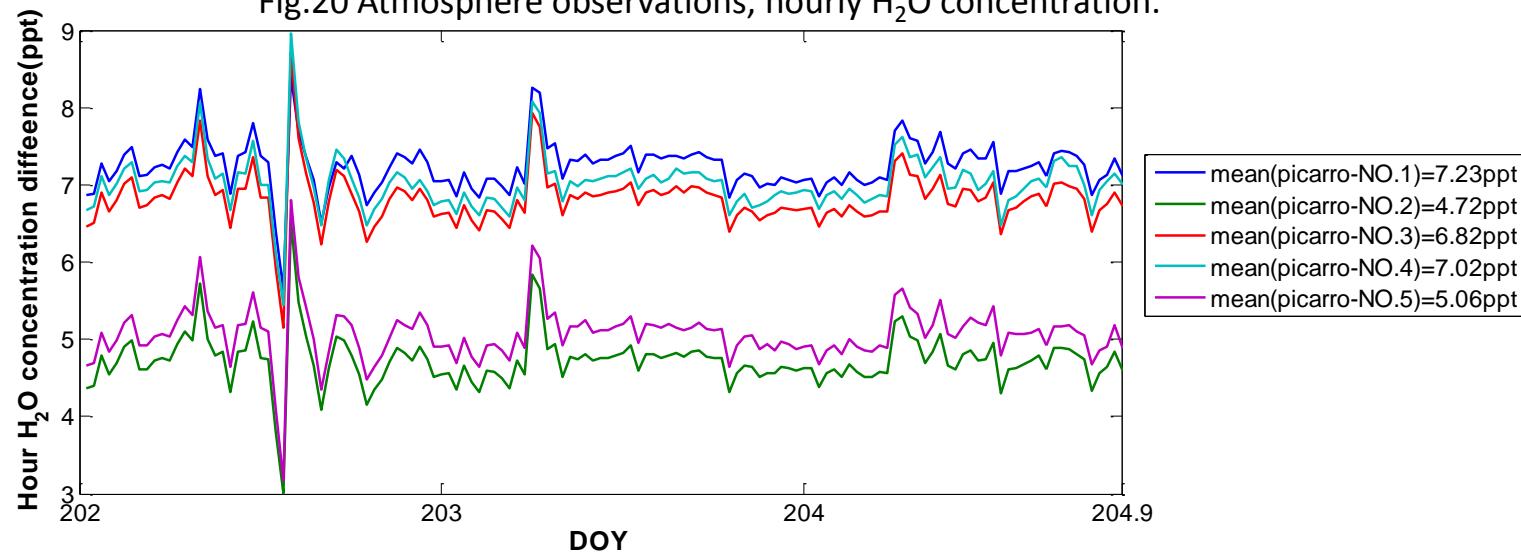


Fig.21 Atmosphere observations, hourly dry  $\text{H}_2\text{O}$  concentration difference.

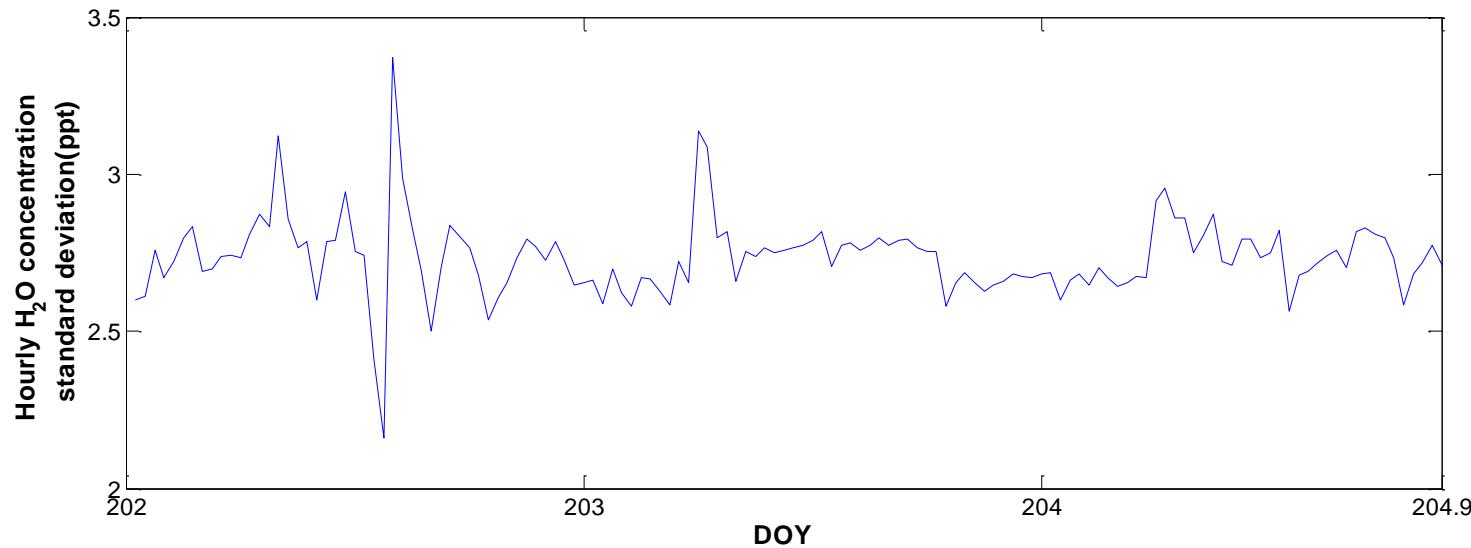


Fig.21 Atmosphere observations ,hourly H<sub>2</sub>O concentration standard deviation.

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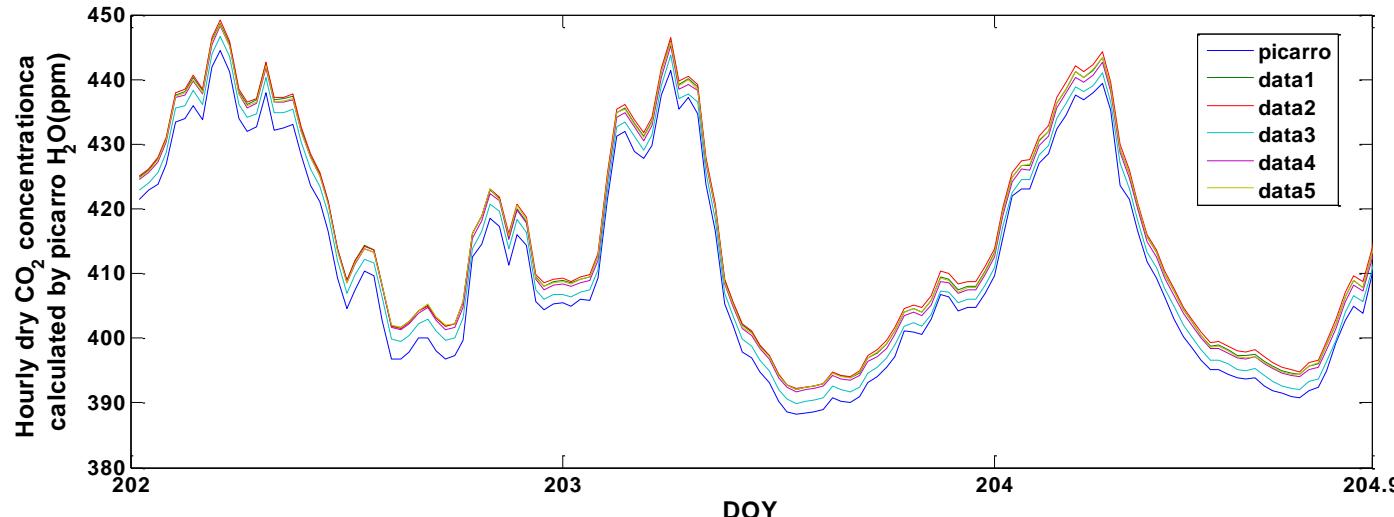


Fig.22 Atmosphere observations, hourly dry  $\text{CO}_2$  concentration calculated by picarro  $\text{H}_2\text{O}$

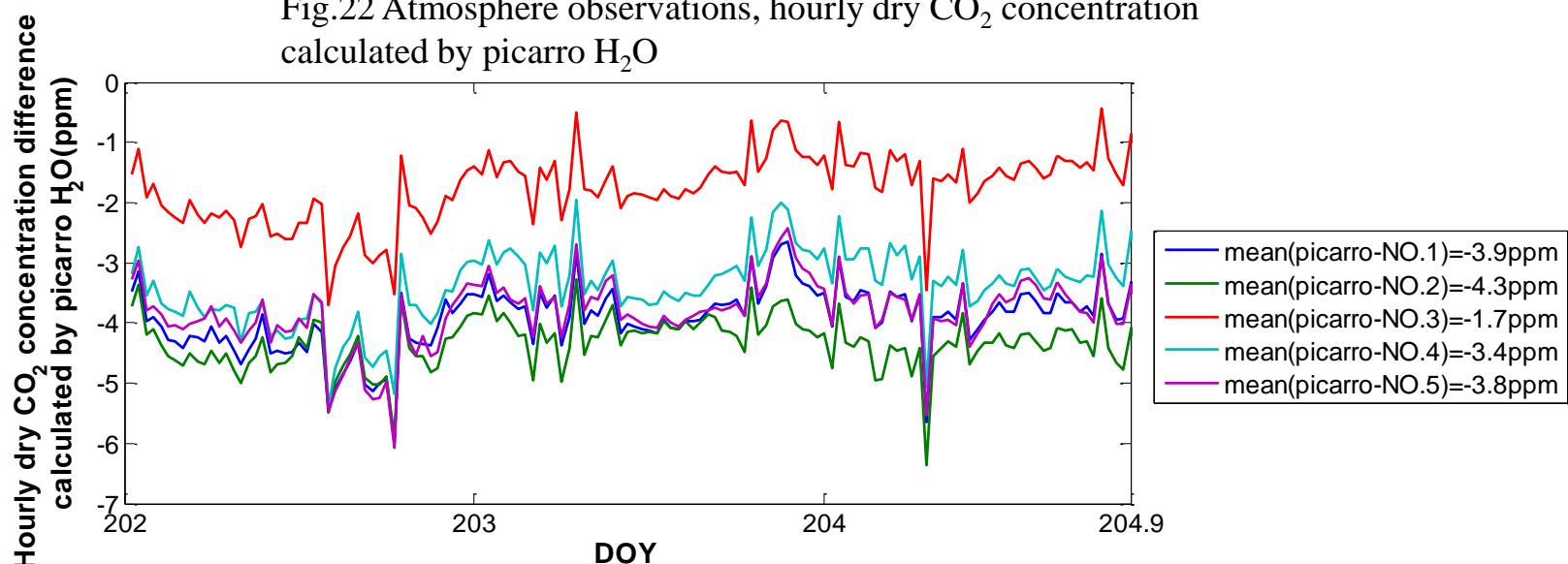


Fig.23 Atmosphere observations, hourly dry  $\text{CO}_2$  concentration difference calculated by picarro  $\text{H}_2\text{O}$ .

## Conclusion

We can take observation with this instrument if someone take care of it.

Calibration everyday when the instrument works, So that we can control the drift of CO<sub>2</sub> in ±3ppm.

Thank you